

Chemistry: Matter Unit Study Guide

6. Identify each image below as an Element, Compound, or Mixture.



Name: _____ Date: _____ Period: _____

Matter Test Study Guide

1. Identify if the following images are of Solid, Liquid, or Gas particles. In the table, describe the characteristics of each state of matter.



State of Matter:

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Shape			
Volume			
Compressibility			

2. What is the difference between a Pure Substance and a Mixture?

3. How can you distinguish if a Pure Substance is an Element or a Compound?

4. Describe the difference between a Homogeneous and Heterogeneous mixture?

5. Identify each item below as an element, compound, homogeneous mixture, or heterogeneous mixture.

Canal with Milk	
Marble (CaCO_3)	
Isopropyl	
Iron (Fe)	

Structure Of Matter Study Guide Chemistry

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Structure Of Matter Study Guide Chemistry:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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Resources in Education, 1994

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