Periodic Table of the Elements

KEY

Atomic Mass → 12.011
Symbol → C
Atomic Number → 6
Electron Configuration → 2-4

← Selected Oxidation States: −4, +2, +4

Relative atomic masses are based on $^{12}C = 12$ (exact)

Note: Numbers in parentheses are mass numbers of the most stable or common isotope.

Period	Group 1	Group 2	Group 3	Group 4	Group 5	Group 6	Group 7	Group 8	Group 9	Group 10	Group 11	Group 12	Group 13	Group 14	Group 15	Group 16	Group 17	Group 18
1	H																	He
2	Li	Be											B	C	N	O	F	Ne
3	Na	Mg											Al	Si	P	S	Cl	Ar
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
6	Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
7	Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Uuq	Uup	Uuh	Uus	Uuo

Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Periodic Table Reference Sheet

Rachel Sandford

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Table of Contents Periodic Table Reference Sheet

1. Understanding the eBook Periodic Table Reference Sheet
    - The Rise of Digital Reading Periodic Table Reference Sheet
    - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Reference Sheet
    - Exploring Different Genres
    - Considering Fiction vs. Non-Fiction
    - Determining Your Reading Goals
3. Choosing the Right eBook Platform
    - Popular eBook Platforms
    - Features to Look for in an Periodic Table Reference Sheet
    - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Reference Sheet
    - Personalized Recommendations

- Periodic Table Reference Sheet User Reviews and Ratings
- Periodic Table Reference Sheet and Bestseller Lists

5. Accessing Periodic Table Reference Sheet Free and Paid eBooks
    - Periodic Table Reference Sheet Public Domain eBooks
    - Periodic Table Reference Sheet eBook Subscription Services
    - Periodic Table Reference Sheet Budget-Friendly Options
6. Navigating Periodic Table Reference Sheet eBook Formats
    - ePub, PDF, MOBI, and More
    - Periodic Table Reference Sheet Compatibility with Devices
    - Periodic Table Reference Sheet Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Periodic Table Reference Sheet
    - Highlighting and Note-Taking Periodic Table Reference Sheet
    - Interactive Elements Periodic Table Reference Sheet
8. Staying Engaged with Periodic Table Reference Sheet
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Periodic Table Reference Sheet
9. Balancing eBooks and Physical Books Periodic Table Reference Sheet
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Periodic Table Reference Sheet
10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Reference Sheet
    - Setting Reading Goals Periodic Table Reference Sheet
    - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Reference Sheet
    - Fact-Checking eBook Content of Periodic Table Reference Sheet

    - Distinguishing Credible Sources
13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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