

AP Chemistry: A Sample Formal Laboratory Report

This paper is designed to help you prepare a chemistry lab report. Keep it in your chemistry notebook. All chemistry lab reports must be written neatly and well organized to receive full credit. Lab reports may be written or typed. It is highly recommended that you use graph or engineering bond paper for written reports.

Laboratory #7: Quantitative Determination of an Empirical Formula

I. Hypothesis: If nitric acid is poured onto tin, a tin oxide will be produced. If we know the initial mass of the tin metal and the mass of the final product, we can determine the empirical formula of the tin oxide product. There should be a whole number ratio between oxygen and tin. *(The hypothesis explains what is to be tested and will be written after reading the entire laboratory worksheet.)*

II. Equipment: *(Non-chemical equipment used in the experiment.)*
evaporating dish
watch glass
stirring rod
burner with ring stand, ring and wire gauze
forceps
beaker
balance

III. Reagents: *(A listing of chemicals used in the experiment with their amounts and any warnings.)*
tin metal (granulated) ~2 g.
5 cm³ (mL) nitric acid (HNO₃) **caution! severe burns**

IV. Procedure: Each step of the procedure must be written here. You may paraphrase and shorten the explanations, but the reader must be able to perform the experiment from these instructions. *(The procedure must be read carefully before the lab begins. Drawings of the experimental setup may be included here. The teacher may make changes to the procedure; make sure that you write any changes down!)*

V. Data: *(If the laboratory report is handwritten use a ruler to draw data tables and graphs! Always include units with all data entries.)*

	Procedure	Trial 1	Trial 2
a	mass of dish, and watch glass	74.14 g	
b	mass of dish, glass and tin	76.20 g	
c	mass of tin =b-a	2.06 g	
d	moles of tin	.0173 mol	
e	mass of dish, glass, and product	76.76 g	
f	mass of oxygen =e-b	.56 g	
g	moles of oxygen	.0350 mol	
h	mole ratio	2.02 : 1	
i	accepted ratio	2 : 1	
j	% error	1.00 %	

Sample Chemistry Laboratory Report

**Government Chemical Laboratories
(W.A.)**



Sample Chemistry Laboratory Report:

Quality Assurance Representative's Guide ,1992 **Combined Compilation of Meat and Poultry Inspection Issuances for ...** ,1991 **Handbook of Quality Assurance for the Analytical Chemistry Laboratory** J. Dux,2013-11-11

xii a second edition might be in order and readily agreed Although the basic principles remain the same discussions with analysts laboratory supervisors and managers indicated many areas where improvements could be made For example new chapters have been added on sampling and quality assurance laboratory facilities and quality assurance and auditing for quality assurance Very little of the first edition has been discarded but many topics have been expanded considerably The chapter on computers has been completely rewritten in view of the rapid changes in that field The chapter in the first edition on planning and organizing for quality assurance has been split into two chapters one on planning for quality assurance and the other on organizing and establishing a quality assurance program and new material on mandated quality assurance programs has been combined with the material on laboratory accreditation Numerous examples especially those involving mathematical calculations have been added at the suggestion of some readers In short this edition is very nearly a new book and I can only hope it is as well received as the first edition

CHAPTER 1 Quality Quality Control and Quality Assurance One of the strongest trends in modern society is the continuing evolution from a manufacturing to a service oriented economy

Science Educator's Guide to Laboratory Assessment Rodney L. Doran,2002 The book opens with an up to date discussion of assessment theory research and uses Then comes a wealth of sample assessment activities in biology chemistry physics and Earth science Keyed to the National Science Education Standards the activities include reproducible task sheets and scoring rubrics All are ideal for helping students reflect on their own learning during science lab

Report of the Chemical Laboratory (Western Australia. Government Mineralogist, Analyst and Chemist) Government Chemical Laboratories (W.A.),1985 **Practical Chemistry Labs** Leonard Saland,1989 Features self contained step by step activities using common materials and covering topics from food chemistry to papermaking and electrochemistry Illustrates the connection between the real world and chemistry concepts such as solutions chemistry acids and bases and more Includes teacher notes quizzes and answers to help monitor student progress

Report United States. Congress. House,1939 **NTP GMM.** ,2005 **Compilation of Meat and Poultry Inspection Issuances** ,1985-11 **Sampling and Analysis of Rain** S.A. Campbell,1981 **Report** National Center for Radiological Health (U.S.). Interlaboratory Technical Advisory Committee,1965 **Teaching Undergraduate Science** Linda C. Hodges,2023-07-03 This book is written for all science or engineering faculty who have ever found themselves baffled and frustrated by their undergraduate students lack of engagement and learning The author an experienced scientist faculty member and educational consultant addresses these issues with the knowledge of faculty interests constraints and day to day concerns in mind Drawing from the research on learning she offers faculty new ways to think about the struggles their science students face She then provides a range of

evidence based teaching strategies that can make the time faculty spend in the classroom more productive and satisfying Linda Hodges reviews the various learning problems endemic to teaching science explains why they are so common and persistent and presents a digest of key ideas and strategies to address them based on the research she has undertaken into the literature on the cognitive sciences and education Recognizing that faculty have different views about teaching different comfort levels with alternative teaching approaches and are often pressed for time Linda Hodges takes these constraints into account by first offering a framework for thinking purposefully about course design and teaching choices and then providing a range of strategies to address very specific teaching barriers whether it be students motivation engagement in class ability to problem solve their reading comprehension or laboratory research or writing skills Except for the first and last chapters the other chapters in this book stand on their own i e can be read in any order and address a specific challenge students have in learning and doing science Each chapter summarizes the research explaining why students struggle and concludes by offering several teaching options categorized by how easy or difficult they are to implement Some for example can work in a large lecture class without a great expenditure of time others may require more preparation and a more adventurous approach to teaching Each strategy is accompanied by a table categorizing its likely impact how much time it will take in class or out and how difficult it will be to implement Like scientific research teaching works best when faculty start with a goal in mind plan an approach building on the literature use well tested methodologies and analyze results for future trials Linda Hodges message is that with such intentional thought and a bit of effort faculty can succeed in helping many more students gain exciting new skills and abilities whether those students are potential scientists or physicians or entrepreneurs Her book serves as a mini compendium of current research as well as a protocol manual a readily accessible guide to the literature the best practices known to date and a framework for thinking about teaching

Applications of Environmental Aquatic Chemistry Eugene R. Weiner, 2008-01-17 Professionals and students who come from disciplines other than chemistry need a concise yet reliable guide that explains key concepts in environmental chemistry from the fundamental science to the necessary calculations for applying them Updated and reorganized *Applications of Environmental Aquatic Chemistry A Practical Guide* Second Editi Embankment Dam Instrumentation Manual Charles L. Bartholomew, 1987 **Bibliography of Scientific and Industrial Reports** ,1947 *Consolidated Annual Report on State and Territorial Public Health Laboratories* ,1975 **Integrated Approach to Coordination Chemistry** Rosemary A. Marusak, Kate Doan, Scott D. Cummings, 2007-03-30 Coordination chemistry is the study of compounds formed between metal ions and other neutral or negatively charged molecules This book offers a series of investigative inorganic laboratories approached through systematic coordination chemistry It not only highlights the key fundamental components of the coordination chemistry field it also exemplifies the historical development of concepts in the field In order to graduate as a chemistry major that fills the requirements of the American Chemical Society a student needs to take a laboratory course in inorganic chemistry Most

professors who teach and inorganic chemistry laboratory prefer to emphasize coordination chemistry rather than attempting to cover all aspects of inorganic chemistry because it keeps the students focused on a cohesive part of inorganic chemistry which has applications in medicine the environment molecular biology organic synthesis and inorganic materials **Annual**

Reports of the Department of Agriculture for the Fiscal Year Ended ... United States. Department of Agriculture, 1909

The Analytical Chemistry Laboratory Companion Michael D. Holloway, 2025-07-15 The Analytical Chemistry Laboratory Companion is essential for both students and professionals as it provides quick clear explanations on critical topics in analytical chemistry equipping you with the statistical tools necessary to ensure accurate and reliable data interpretation The Analytical Chemistry Laboratory Companion serves as a reference guide for students and professionals alike who need quick explanations on specific topics laboratory operations the structure of designing experiments and the use of statistics to gain increased accuracy precision repeatability and reproducibility of data This volume will also provide in depth and advanced studies and build the necessary background knowledge for success in the field This companion provides a concise examination of the various analytical tools used for chemistry and defines basic analytical instrument principles techniques and applications in addition to exploring statistical tools useful in data interpretation test result reporting and common root causes for faulty data with suggested remedies The introduction provides a concise guide on foundational topics such as developing standard operating procedures laboratory safety instrumental analytical methods and common statistical tools useful for data interpretation This companion covers both wet chemical and instrumental analysis including their principles applications and pitfalls The Analytical Chemistry Laboratory Companion is a must have comprehensive guide in the field of analytical chemistry **Computerization and Data Management in the Metals Analysis**

Laboratory Mary Ann Worthington, Norma L. Bottone, 1988

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