# Reaction Quotient

How do you predict which direction a reaction will proceed to reach equilibrium?

## Why?

When a reaction reaches equilibrium there must be some non-negligible amount of every species in the reaction, otherwise the reaction cannot react in both directions. Knowing this, it is very easy to predict which direction a reaction will go to reach equilibrium when one of the components of the reaction has an initial concentration of zero. Many of the problems you have worked with thus far have some component at zero concentration, but real life does not work that way. Most of the time, the reaction in question has some measureable amount of every species. Deciding which way the reaction will go to reach equilibrium then becomes more challenging.

## Model 1 - A Theoretical Equilibrium

Trial I	$A(g) + B(g) \longleftrightarrow C(g)$		
Initial	1,000 M4	1.000 M	1.000 M
Change	<		
Equilibrium	1.464 M	1.464 M	0.536 M

Trial 2	$A(g) + B(g) \longleftrightarrow C(g)$		
Initial	2.000 M	0.500 M	0.500 M
Cihange			
Equilibrium	2.150 M	0.650 M	0.350 M

Trial 3	$A(g) + B(g) \longleftrightarrow C(g)$		
Inicial	1.000 M	0.500 M	1.500 M
Change			
Equilibrium	1.864 M	1.364 M	0.636 M

Trial 4	$A(g) + B(g) \longleftrightarrow C(g)$		
Unitial	1.600 M	1,000 M	0.400 M
Change			
Equilibrium	1,600 M	1.000 M	0.400 M

Trial 5	$A(g) + B(g) \longleftrightarrow C(g)$		
Initial	1.400 M	1.200 M	0.400 M
Change			
Equilibrium	1.388 M	1.188 M	0.412 M

Trial 6	$A(g) + B(g) \longleftrightarrow C(g)$		
Unitial	0.750 M	2.000 M	0.250 M
Change			
Equilibrium	0.675 M	1.925 M	0.325 M

- 1. Examine Model 1.
  - at. Write the theoretical chemical reaction that is used in the trials of Model 1.
  - & If 0.50 M of reactant A reacts, predict the change in concentration of B and C.,
- 2. What variables were changed in the different trials shown in Model 1?

# **Pogil Activities For Ap Chemistry**

**David Hanson** 

#### **Pogil Activities For Ap Chemistry:**

**POGIL** Shawn R. Simonson, 2023-07-03 Process Oriented Guided Inquiry Learning POGIL is a pedagogy that is based on research on how people learn and has been shown to lead to better student outcomes in many contexts and in a variety of academic disciplines Beyond facilitating students mastery of a discipline it promotes vital educational outcomes such as communication skills and critical thinking Its active international community of practitioners provides accessible educational development and support for anyone developing related courses Having started as a process developed by a group of chemistry professors focused on helping their students better grasp the concepts of general chemistry The POGIL Project has grown into a dynamic organization of committed instructors who help each other transform classrooms and improve student success develop curricular materials to assist this process conduct research expanding what is known about learning and teaching and provide professional development and collegiality from elementary teachers to college professors As a pedagogy it has been shown to be effective in a variety of content areas and at different educational levels This is an introduction to the process and the community Every POGIL classroom is different and is a reflection of the uniqueness of the particular context the institution department physical space student body and instructor but follows a common structure in which students work cooperatively in self managed small groups of three or four The group work is focused on activities that are carefully designed and scaffolded to enable students to develop important concepts or to deepen and refine their understanding of those ideas or concepts for themselves based entirely on data provided in class not on prior reading of the textbook or other introduction to the topic The learning environment is structured to support the development of process skills such as teamwork effective communication information processing problem solving and critical thinking The instructor s role is to facilitate the development of student concepts and process skills not to simply deliver content to the students The first part of this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about The POGIL Project POGIL Activities for AP\* Chemistry Flinn Scientific, 2014 Chemistry Richard S. Moog, John J. Farrell, 2017-06-26 In the newly updated 7th Edition Chemistry A Guided Inquiry continues to follow the underlying principles developed by years of extensive research on how students learn and draws on testing by those using the POGIL methodology This text follows the principles of inquiry based learning and correspondingly emphasizes underlying chemistry concepts and the reasoning behind them This text provides an approach that follows modern cognitive learning principles by having students learn how to create knowledge based on experimental data and how to test that knowledge

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