- 11) All the other substances besides SO₂, HCl, & NH₃ are ionic compounds which means they are what physical state? solids
- 12) What conclusions can you draw about the effects of temperature on the solubility for gases compared to that of solids?

As temperature increases, the solubility of gases decreases but the solubility of solids increases.

- 13) What substances have saturation points that are the most vulnerable to fluctuations in temperature? KNO₃, NaNO₃ & NH₃
- 14) What substance has a saturation point that is most resistant to changes in temperature? NaCl
- 15) Suppose 12 grams of SO₂ is dissolved in water at 1°C. The water is then heated. At what temperature will the SO₂ bubble out of the solution? About 12°C
- 16) Suppose 40 grams of KCIO: is dissolved in water at 95°C. The water is then cooled. At what temperature will the KCIO: precipitate out of the solution? About 78°C
- 17) Suppose 40 grams of NH₂Cl is dissolved in water at 85°C. The water is then cooled to 50°C. Will a precipitate form? Explain your answer. No, the saturation point of NH₂Cl at 50°C is about 51 grams so no solid precipitate will form.

- 18) 90 grams of KNOs is placed in 100ml of water # 40°C...
 - a) Will the solution be saturated? If the answer is yes, how much (if any) KNOs will remain undissolved at the bottom of the container? If the answer is no, how much more KNOs could be dissolved? Yes, about 28 grams will remain undissolved.
 - b) If the solution is heated to 70°C will it be saturated? If the answer is yes, how much (if any) KNOs will be undissolved at the bottom of the container? If the answer is no, how much more KNOs could be dissolved? No, about 45 more grams of KNOs could be dissolved at this temperature.
 - c) If the solution is cooled to 20°C will it be saturated? If the answer is yes, how much (if any) KNOs will be undissolved at the bottom of the container? If the answer is no, how much more KNOs could be dissolved? Yes, about 58 grams will be re-crystalized or undissolved.
- 19) 90 grams of NaNOs is placed in 100ml of water ⊕ 40°C...
 - a) Will the solution be saturated? If the answer is yes, how much (if any) NaNOs will remain undissolved at the bottom of the container? If the answer is no, how much more NaNOs could be dissolved? No, about 15 more grams could be dissolved.
 - b) If the solution is heated to 70°C will it be saturated? If the answer is yes, how much (if any) NaNOs will be undissolved at the bottom of the container? If the answer is no, how much more NaNOs could be dissolved? No, about 45 more grams of NaNOs could be dissolved.
 - c) If the solution is cooled to 20°C will it be saturated? If the answer is yes, how much (if any) NaNOs will be undissolved at the bottom of the container? If the answer is no, how much more NaNOs could be dissolved? Yes, about 3 grams will be re-crystalized.

Pogil Solubility Curve Answer Key

S Marginson

Pogil Solubility Curve Answer Key:

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