Skill Practice 19



1. Write the symbol and charges for the following ions:

Example: calcium v Ca

- A/I pfluospfleide m
- B) magnesises w _____ C) rehidises w

- D) fluoride = E) aluminum = F) sulfide =
- 2. Each of the following formulas is written incorrectly. Please rewrite them correctly. Hyamphr: Ca₂Cl = CaCl₂

- A) BaS = ______ B) RbN = ____ C) LbC = ____
- D) Al₂N₁ = _____ F) Mg/Br₂ = ____ F) O₂Al₂ = ____

- 3. Write the formulas for the compound formed by the combination of the following pairs of atoms. Example: nitrogen and magnesium a Ma-N-

A) redison and suffer = _______ B) alaminum and obtoring =

C) phosphorus and calcium = D) burium and oxygen =

 For each of the formulas you wrote in question 3 above, give the names for the compounds. Example: Mg-N_c w magnesium mbride

C) D)

5. In the blanks above each column, write the charge of the ions formed by atoms in that column. Note: group IA is done for you.



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