

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____

Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends, (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	152	Sodium - 11	190	Potassium - 19	243	Rubidium - 27	265
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 28	219
Boron - 5	87	Aluminum - 13	118	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 52	125	Tin - 50	140
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	115

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 27	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 28	550

Atomic Radius



Trends In Periodic Table Graph Answers

Wolfgang Guggemos

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Trends In Periodic Table Graph Answers:

Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. WileyPLUS sold separately from text.

Science Units for Grades 9-12 Randy L. Bell, Joe Garofalo, 2005 Sample topics include cell division, virtual dissection, earthquake modeling, the Doppler Effect, and more.

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The Science Teachers Bulletin, 1989

Principles of Epidemiology Workbook: Exercises and Activities Ray M. Merrill, 2010-09-15 This workbook was written for students of epidemiology and serves as a supplement to any one of several introductory text books in epidemiology. Each chapter is divided into an introduction, a series of questions and detailed responses, and a series of Homework questions. At the end of each chapter is a table with a list of selected epidemiology text books with accompanying chapters in those books that the workbook chapter may supplement. The general learning outcomes (LOs) for this workbook are:

1. Become familiar with basic concepts and definitions commonly used in epidemiology.
2. Define a public health problem.
3. Identify appropriate uses and limitations of data and research design strategies for solving public health problems.
4. Make relevant inferences from quantitative and qualitative data.
5. Distinguish between statistical association and cause-effect relationships.
6. Measure and describe patterns of disease incidence, prevalence, and mortality.
7. Identify environmental factors and behaviors associated with health-related states or events.
8. Be familiar with the steps for investigating disease outbreaks.
9. Identify, calculate, and interpret common indices used in identifying the health status.
10. Evaluate program effectiveness.
11. Critically assess epidemiological research.
12. Be able to communicate health findings.

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NEST : National Entrance Screening Test (General Aptitude, Physics, Chemistry, Mathematics, Biology) - 10 Mock Tests (Solved Questions) with Free Access to Online Tests EduGorilla Prep Experts, 2020-12-28 National Entrance Screening Test (NEST) is conducted jointly by National Institute of Science Education and Research (NISER) Bhubaneswar and Mumbai University. The candidates that are

able to successfully qualify the exams will be offered admission in the above mentioned prestigious universities The NEST Exam is conducted annually to offer admission for Integrated MSc programme in Biology Chemistry Mathematics Physics

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science teachers of grades 5-12. The book offers a dynamic collection of classroom ready lessons, projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life. **Cambridge IGCSE® Physical Science Chemistry Workbook** Richard Harwood, Ian Lodge, 2017-02-16. Cambridge IGCSE Physical Science resources tailored to the 0652 syllabus for first examination in 2019 and all components of the series are endorsed by Cambridge International Examinations. This Chemistry Workbook is tailored to the Cambridge IGCSE Physical Science 0652 syllabus for first examination in 2019 and is endorsed for learner support by Cambridge International Examinations. The workbook covers both the Core and the Supplement material. Developing students' scientific skills, the workbook exercises are complemented by self-assessment checklists to help students evaluate their work as they go. Answers are provided at the back of the book.

Essential AS Chemistry for OCR Ted Lister, Janet Renshaw, 2004. Essential AS Chemistry for OCR provides clear progression with challenging material for in-depth learning and understanding. Written by the best-selling authors of *New Understanding Chemistry*, these texts have been written in simple, easy-to-understand language and each double-page spread is designed in a contemporary manner. Fully networkable and editable Teacher Support CD-ROMs are also available for this series; they contain worksheets, marking schemes and practical help. **Johnstone Triangle** Norman Reid, 2021-01-18. Chemistry is often seen as a difficult subject to understand. This book focusses on the triangle model that Alex H. Johnstone developed in the early 1980s. Originally conceived in the context of making chemistry more accessible to a wider range of learners, the model has been applied in almost every area of education in chemistry at all stages of learning. In looking at why chemistry is difficult, there are two central questions: Firstly, does the problem relate to the nature of chemistry, and secondly, does it relate to the way humans gain understanding? Both were found to be important, and the answers to the two questions were found to be connected. The triangle model arose from sustained research into human learning. The central finding from research is the critical role of working memory, and the model rationalises so much evidence from chemistry education research as well as the repeated experiences of teachers of chemistry at all levels. In order to understand chemistry, it is essential to develop sound mental models of molecular reality. It generates major implications for the way a chemistry curriculum should be constructed and the processes of teaching and learning in chemistry when the goal is focussed on understanding the key ideas. Some of these implications are developed, and pointers offered to more successful ways forward. The power of the Johnstone Triangle lies in the way it offers clear directions for all involved in chemistry education. It is hoped that this book will prove helpful to all involved in sharing the exciting story of the way humans have come to understand the molecular world, one of the great examples of great human endeavour. **Essential Laboratory Mathematics** Catherine W. Johnson, Daniel L. Timmons, Pamela E. Hall, 2009-12-03. This hands-on manual with pedagogical features that draw the learner into the content offers clear and complete coverage of the mathematical topics most often used in today's clinical and medical laboratories. Furthermore, it provides a solid foundation for subsequent courses in the laboratory sciences.

The first two chapters present a review of basic mathematical concepts The remainder of the book provides students with a realistic means to build on previously learned concepts both mathematical and scientific to refine their mathematical skills and to gauge their mastery of those skills Outstanding features Each chapter opens with an outline objectives and key terms Key terms highlighted within the text are listed and defined in the glossary Margin problems and practice problem sets provide the chance to gain immediate proficiency Laboratory exercises and review problems allow students to apply what they ve learned and assess their understanding and progress A special calculator icon signals explanations of calculator use for a particular mathematical function Study hints Keys to Success offer practical suggestions and guidance for maximizing achievement The workbook design enables users to solve problems and take notes directly on the pages [Sif Chemistry NI Twb 2e](#) Rex M. Heyworth,2007

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