Discriminant

Find the discriminant of each quadratic equation.

1)
$$6x^2 + 3x = 7$$

$$-2\alpha = -5\alpha^2$$

3)
$$9 = p(4+p) + 5$$

4)
$$2t - 7t^2 = -2$$

5)
$$(y + 1)(y - 8) = 0$$

6)
$$-10s + 2 = 7s^2$$

7)
$$r^2 + 3r + 5 = 0$$

8)
$$n^2 = -4$$

9)
$$9u^2 + 6u - 2 = 0$$

10)
$$5v + 2v^2 = 1 + v$$

11)
$$-8 = 3(w^2 - 1)$$

12)
$$q = 7q^2 + 2$$

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web characteristics physical acids would look solid liquid or in the form of gas it would also have a sour taste bases would feel slippery and solid in nature except for ammonia

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web log h 10 4 05 8 91 x 10 5 mol dm 3 log pka ph 3 75 4 05 0 30 10 0 30 0 50 and hcooh 0 1 so hcoo 0 200 mol dm 3 amount

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web only the better candidates could write successive ionisation equations in part c while part d about the ph of equimolar strong and weak acid solutions was poorly answered

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web a substance may be assigned to one our four conceivable categories it may be an acid or a base but in addition it may be both an acid and a base or it may be neither an acid nor

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web to detect the presence of acids and bases micheal adds a few drops of phenolphthalein indicator to two colorless solutions a and b he observes the following changes in the

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web nov 5 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like zinc magnesium aluminium or iron acids have a low ph

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web section 2 ph of strong acids number of protons released monoprotic acid acid that releases one h ion per molecule e g hcl hydrochloric acid hno 3 nitric acid ch

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web jan 10 2021 s14 3 21 strong electrolytes are 100 ionized and as long as the component ions are neither weak acids nor weak bases the ionic species present result

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web problem 1 which of the following are not acids ch3cooh co2 hno2 hcooh cc4 c h 3 c o o h c o 2 h n o 2 h c o o h c c l 4 shazia naz numerade educator 02 23

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web in grade 11 learners were introduced to a lot of the concepts that are expanded on in this chapter these include acid and base models and definitions conjugate acid base

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web which is the conjugate base in the following reaction multiple choice question 13 which of the following compounds is the strongest acid multiple choice question 14 which

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web nov 13 2022 suggest simple tests you could carry out to determine if an unknown substance is an acid or a base state the chemical definitions of an acid and a base in

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web 2 at what ph or range of ph values is a solution considered strongly acidic mildly acidic neutral mildly basic and strongly basic provide a brief explanation

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web 11 why is hcl a strong acid and hclo a weak acid 12 why are hcl and hclo 4 both strong acids 13 for each of the reactions below classify the reactants as an acid or a

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web answer i lemon and orange ii tomatoes and guava iii sour milk curd iv tamarind question 18 a student detected the ph of four unknown solution a b c and d as