

## The Periodic Table

H																	He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac	Rf	Ha	Sg	Uns	Uno	Une									
		Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
		Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		

1. The elements in periods 2 through 7 that are to the left of the zigzagged line are metals. List five metals.  
\_\_\_\_\_
2. The elements in periods 1 through 6 that are to the right of the zigzagged line are nonmetals. List five nonmetals.  
\_\_\_\_\_
3. The elements that border either side of the zigzagged line are metalloids. Metalloids have properties of both metals and nonmetals. Name two metalloids.  
\_\_\_\_\_
4. The alkali metals are in group 1. They have only one electron in their outer shell. They are very reactive and have a low melting point. Shade the alkali metals blue.
5. The noble gases are found in group 18. Their outer energy level is filled, and they are very nonreactive, colorless gases. Shade the noble gases red.
6. Halogens have seven electrons in the outer shell. They are located in group 17 on the periodic table. Halogens react with metals to form salts. Shade the halogens green.
7. The transition elements are found in groups 3 through 12 and periods 4 through 7. These elements have either one or two electrons in the outer energy level. They are often used to form alloys because they are hard and have high melting points. Shade the transition elements orange.
8. Alkaline earth metals are located in group 2. They have two electrons in their outer energy level. Shade the alkaline metals purple.

# The Periodic Table Worksheet Answer Key Section 2

**Carson-Dellosa Publishing**



## **The Periodic Table Worksheet Answer Key Section 2:**

*Oxford Smart Activate Chemistry Teacher Handbook (Ebook)* Alyssa Fox-Charles, 2025-06-05 Oxford Smart Activate Chemistry Teacher Handbook Ebook has high aspirations for all budding chemists at KS3 Building on what has been learned at KS2 this handbook helps teachers to plan and deliver lessons that immerse learners in the world of chemistry while developing key knowledge and skills towards GCSE Providing support for all teachers specialists and non subject specialists this handbook contains practical suggestions to reactivate prior knowledge trigger student interest and reflect on learning and progress Links between topics sciences and the wider KS3 curriculum are clearly identified Informed by up to date educational research and tried and tested by Pioneer Schools UK to ensure that the teacher guidance is relevant impactful and promoting current pedagogical practice Oxford Smart Activate is the next evolution of the best selling Activate series from editor and curriculum expert Andrew Chandler Grevatt

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**The School Science Review**, 2003

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