



The Railfan Chronicles

Ann Arbor Railroad
Volume 2 1981 to 2000

Byron Babbish

The Railfan Chronicles Ann Arbor Railroad Volume 2 1981 To

Huangqi Zhang



The Railfan Chronicles Ann Arbor Railroad Volume 2 1981 To :

The Railfan Chronicles, Ann Arbor Railroad, Volume 2, 1981 To 2000 Byron Babbish, 2014-11-07 The Railfan Chronicles is a series of books about different railroads during the period from 1975 to 2000 This was a period of great change in the railroad industry with mergers of large railroads and creation of regional and short line railroads to operate unwanted lines of the Class One railroads These books will cover this period in photographs taken by the author This is Volume 2 of our coverage of the Ann Arbor Railroad looking at its operations from Toledo Ohio across Michigan between the years 1981 and 2000 During this period of time the State of Michigan owned the Annie and its operator changed from Michigan Interstate Railroad Company to Tuscola Saginaw Bay Railway Its northern end was abandoned and the Lake Michigan Carferry operations stopped The evolution of the Ann Arbor Railroad during this period is shown in both color and black and white photographs Chronicle ,1983 **Railfan & Railroad** ,1983-11

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 40 g of H_2O molar mass of H_2O is 18 gmol^{-1} and molar mass of CH_3OH is 32 gmol^{-1} solution moles of CH_3OH $4.5 / 32 = 0.14$
mole moles of H_2O $40 / 18 = 2.2$ moles therefore according to the equation

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time the total pressure $p_a = x_a p_t$ mole fraction $x_a = \frac{\text{moles } a}{\text{moles total}}$ $\sum_{i=1}^n x_i = 1$ mole fraction of substance
 a is often denoted by x_a and

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0 2885 its mole percent would be 28 85

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