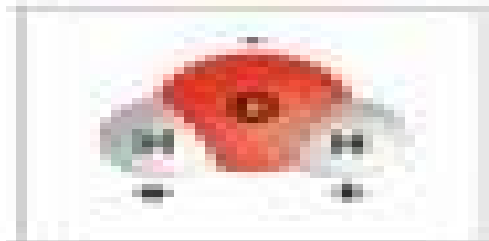


Properties of Water Guided Reading Questions

Directions: Read the Water Properties document. As you read, complete the following guided reading notes.

1. Water is kept together by the two types: hydrogen of the bonds and the charge.



2. Water is "sticky" because cohesion (molecule charged attract) and water has a high surface tension.

3. Why is water called the universal solvent?

Because it dissolves more substances than any other liquid.

4. Water is the only natural substance found both 3 states:
 a. solid, liquid, and gas (none of the temperatures normally found on Earth).

5. Complete the following chart.

Water Property	$^{\circ}\text{F}$	$^{\circ}\text{C}$
freezing point	32	0
boiling point	212	100

6. Why does ice float on liquid water?
water expands so that its solid form does not sink down from the liquid phase.

7. Why is water such a solvent at high specific heat levels?
The hydrogen bonds water used when it's a lot of heat before it begins to get hot.

8. Complete the following chart.

The Properties Of Water Guided Reading Study Answer Key

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