



Name:

Date:

Student Exploration: Covalent Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers?

Flora can give Frank her blue marker and Frank can give her her red marker.

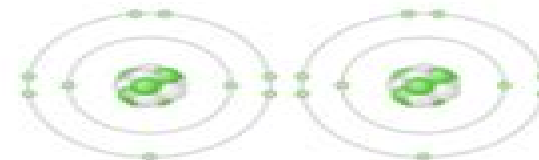
2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

They could combine their sets of markers to have one shared set of markers.

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** (▶) to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have?

2. Click **Pause** (⏸). Drag a valence electron from the left atom to the right atom. Click **Play**.

What happens?

A ring forms around the atoms bonding them together to allow them to rotate in unison.

3. Click **Pause**, drag a valence electron from the right atom to the left, and then click **Play**.

The Nature Of Covalent Bonding 8 2 Answer Key

Arihant Experts



The Nature Of Covalent Bonding 8 2 Answer Key:

NDA / NA General Knowledge (GK) Study Notes | National Defence Academy, Naval Academy Defence Entrance Exam - Theory and Practice Tests for Complete Preparation , **The Nature of Matter Gr. 5-8** , CERAMIC MATERIALS AND PROCESSING - PHASE EQUILIBRIA - PHYSICAL MODELLING IN ELECTRON MICROSCOPY Prof. Dr. ZIYA ENGİN ERKMEN,2019-09-15 There were two main driving forces in my decision for preparing a question answer book covering all the courses given by myself during the past 10 years in my academic career first argument is that there exists a good amount of original questions in the exams and their corresponding answers next my expectation is that probably fewer time would be spent in preparing such a book where the questions and answers are all ready prepared thereupon In this country most of the undergraduate students do not necessarily work on the course they attend in the same day instead they prefer to start preparing their exams 2 or 3 days before In these circumstances last minute students may usually prefer working on the passed questions for the tomorrow s exam However this method of preparation may lead unwanted consequences such that students mostly do not find same questions unconsciously learnt by heart in the exams In order to increase working efficiency and consequently reach to the maximum performance in exams will only be possible if students do not miss any lectures given by staff in addition to be an active participant during the lecture and daily work on the given homeproblems are the basic requirements for a full success In concluding this book will give a chance to the well prepared students to make a quick rehearsal before the exams for obtaining best results after the exams

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- Chapter 2: Essential Elements of The Nature Of Covalent Bonding 8 2 Answer Key
- Chapter 3: The Nature Of Covalent Bonding 8 2 Answer Key in Everyday Life
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- Chapter 5: Conclusion

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




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