

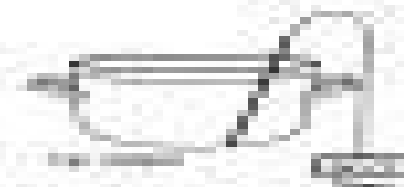
Calorimetry

What is calorimetry? How does it work? What is its purpose?

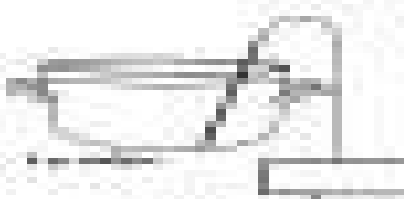
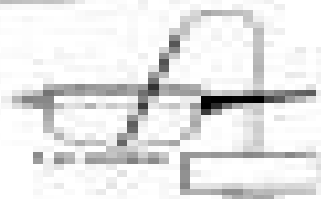
When a substance is heated, the temperature of that substance increases. Will the temperature of every other substance in the same container increase? Will the temperature of every substance in the same container increase by different amounts at the same temperature? Is this possible? How will you know? How much, temperature, heat energy, and the type of substance are related.

Model 1 - A Plot of Mass

Temperature



The temperature of Substance A is 20°C at 0 minutes.



1. In Model 1, which substance has the same amount of heat? Substance A

2. Compare the graphs in Model 1.

a. How much heat energy was added to the substance? 100 J

b. How much heat energy was added to the substance? 100 J

c. In Model 1, which substance has the same amount of heat? Substance A

Why? Because the temperature of Substance A is 20°C at 0 minutes.

Compare the graphs in Model 1. Substance A has the same amount of heat as Substance B.

Substance A

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