

Chem 105 Practice Exam Chapters 6, 5, 7 and 8 OpenStax

1) Of the following, which atom has the smallest atomic radius?
 A) Mg B) Sr C) Te D) S 14) _____

2) Which is the longest bond?
 A) N-N
 B) N=N
 C) N≡N
 D) All three bond lengths should be about the same. 17) _____

3) Of the following elements, which has the lowest electronegativity?
 A) Ca B) Mg C) Cl D) Br 18) _____

4) Which molecule contains the most polar bond?
 A) CF₄ B) CO₂ C) CH₄ D) CCl₄ 19) _____

5) In the most acceptable Lewis structure for carbonyl fluoride, COF₂ the central atom is
 A) O, which is doubly-bonded to C.
 B) C, which is singly-bonded to O.
 C) O, which is singly-bonded to C.
 D) C, which is doubly-bonded to O. 20) _____

Draw Lewis structure here. (2 pts)

6) Which element can accommodate more than eight electrons in its valence shell?
 A) F B) He C) C D) O 21) _____

7) How many resonance structures are present in the Lewis structure of CO₃²⁻?
 A) five B) two C) four D) three 22) _____

Draw Lewis structure here for one resonance form. (2 pts)

8) Assign formal charges to all atoms in the following resonance form for HNO₂. 23) _____



Principles Of Chemistry 103 Lab Manual Answers

Jo Allan Beran



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