

CHAPTER 4

STUDY GUIDE

The Structure of the Atom

Section 4.1 Early Ideas About Matter

In your textbook, read about the philosophers, John Dalton, and defining the atom.

For each statement below, write *true* or *false*.

- _____ 1. Ancient philosophers regularly performed controlled experiments.
- _____ 2. Philosophers formulated explanations about the nature of matter based on their own experiences.
- _____ 3. Both Democritus and Dalton suggested that matter is made up of atoms.
- _____ 4. Dalton's atomic theory stated that atoms separate, combine, or rearrange in chemical reactions.
- _____ 5. Dalton's atomic theory stated that matter is mostly empty space.
- _____ 6. Dalton was correct in thinking that atoms could not be divided into smaller particles.
- _____ 7. Dalton's atomic theory stated that atoms of different elements combine in simple whole-number ratios to form compounds.
- _____ 8. Dalton thought that all atoms of a specific element have the same mass.
- _____ 9. Democritus proposed that atoms are held together by chemical bonds, but no one believed him.
- _____ 10. Dalton's atomic theory was based on careful measurements and extensive research.

Structure Of The Atom Study Guide Answer

Julie R. Frentrup



Structure Of The Atom Study Guide Answer:

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Study Guide/Selected Solutions Manual Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter

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Body Bruce Wingerd, Patty Bostwick Taylor, 2013-02-01 This Student Notebook and Study Guide the ideal companion to Bruce Wingerd's The Human Body reinvents the traditional study guide by giving students a tool to help grasp information in class and reinforce learning outside of class Too often students struggle to both learn the concepts presented and simultaneously record crucial information The Student Notebook and Study Guide provides a structure for recording in class material that parallels the text's concept presentation and includes supplemental questions and activities for assignment outside of the classroom A complete answer guide for both the in class and out of class materials is available online

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