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## Student Exploration: Limiting Reactants

**Vocabulary:** chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

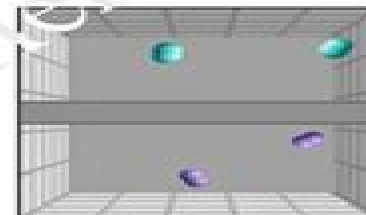
Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

- How many total hot dogs can you make? 30
- Which ingredient limited the number of hot dogs you could make? buns
- Which ingredient will you have leftovers of? hotdogs

### Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the Limiting Reactants Gizmo™, you can determine which reactant is limiting in various scenarios.



To begin, make sure  $\text{H}_2 + \text{O}_2$  becomes  $\text{H}_2\text{O}$  is selected. The small "2" in  $\text{H}_2$ ,  $\text{O}_2$ , and  $\text{H}_2\text{O}$  is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

- Use the sliders to set the number of  $\text{O}_2$  molecules and  $\text{H}_2$  molecules to two.
  - How many hydrogen molecules ( $\text{H}_2$ ) are there? 2 Hydrogen atoms? 4
  - How many oxygen molecules ( $\text{O}_2$ ) are there? 2 Oxygen atoms? 4
- How many  $\text{H}_2\text{O}$  molecules do you think will form when these four molecules react? 2
- Click Play (▶). How many  $\text{H}_2\text{O}$  molecules actually formed? 3



# Student Exploration Limiting Reactants Explore Learning Teacher Key

**Evan Feldman,Ari Contzius**



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