



Name: Tra Nguyen

Date: 11/12/2020

Student Exploration: Electron Configuration

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: atomic number, atomic radius, Aufbau principle, chemical family, diagonal rule, electron configuration, Hund's rule, orbital, Pauli exclusion principle, period, shell, spin, subshell

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Elvis Perkins, a rather shy fellow, is getting on the bus shown at right. Which seat do you think he will probably sit in? Mark this seat with an "E."

Elvis Perkins will most likely sit on the seat on the upper left that is empty with no passengers.

2. Marta Warren gets on the bus after Elvis. She is tired after a long day at work. Where do you think she will sit? Mark this seat with an "M."

Marta Warren will most likely sit next to the door with the white and black fellow. Since she is tired and will want to not have to walk all the way down the rows when it is her time to exit.

3. In your experience, do strangers getting on a bus like to sit with other people if there is an empty seat available? _____

No, people tend to sit and give others space if there is an empty seat available on a bus.



Gizmo Warm-up

Just like passengers getting on a bus, electrons orbit the nuclei of atoms in particular patterns. You will discover these patterns (and how electrons sometimes act like passengers boarding a bus) with the *Electron Configuration Gizmo*.

To begin, check that **Lithium** is selected on the PERIODIC TABLE tab.

1. The **atomic number** is equal to the number of protons in an atom.

How many protons are in a lithium atom? _____

There are 7 protons in a Lithium atom.



| | | | | | |
|----|----|----|----|----|----|
| 3 | 4 | 5 | 6 | 7 | 8 |
| Sc | Ti | V | Cr | Mn | Fe |
| Y | Zr | Nb | Mo | Tc | Ru |
| Lr | Hf | Ta | W | Re | Os |

2. A neutral atom has the same number of electrons and protons.
How many electrons are in a neutral lithium atom? _____

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Table of Contents Student Exploration Electron Configuration Gizmo Answer Key

1. Understanding the eBook Student Exploration Electron Configuration Gizmo Answer Key
 - The Rise of Digital Reading Student Exploration Electron Configuration Gizmo Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Student Exploration Electron Configuration Gizmo Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Student Exploration Electron Configuration Gizmo Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Student Exploration Electron Configuration Gizmo Answer Key
 - Personalized Recommendations
 - Student Exploration Electron Configuration Gizmo Answer Key User Reviews and Ratings
 - Student Exploration Electron Configuration Gizmo Answer Key and Bestseller Lists
5. Accessing Student Exploration Electron Configuration Gizmo Answer Key Free and Paid eBooks
 - Student Exploration Electron Configuration Gizmo Answer Key Public Domain eBooks
 - Student Exploration Electron Configuration Gizmo Answer Key eBook Subscription Services
 - Student Exploration Electron Configuration Gizmo Answer Key Budget-Friendly Options
6. Navigating Student Exploration Electron Configuration Gizmo Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Student Exploration Electron Configuration Gizmo Answer Key Compatibility with Devices
 - Student Exploration Electron Configuration Gizmo Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Student Exploration Electron Configuration Gizmo Answer Key
 - Highlighting and Note-Taking Student Exploration Electron Configuration Gizmo Answer Key
 - Interactive Elements Student Exploration Electron Configuration Gizmo Answer Key

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8. Staying Engaged with Student Exploration Electron Configuration Gizmo Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Student Exploration Electron Configuration Gizmo Answer Key
9. Balancing eBooks and Physical Books Student Exploration Electron Configuration Gizmo Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Student Exploration Electron Configuration Gizmo Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Student Exploration Electron Configuration Gizmo Answer Key
 - Setting Reading Goals Student Exploration Electron Configuration Gizmo Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Student Exploration Electron Configuration Gizmo Answer Key
 - Fact-Checking eBook Content of Student Exploration Electron Configuration Gizmo Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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