



# Chemistry

## Review and Study Guide

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### Analyzing Data

Study Guide: Units and Measurements, Dimensional Analysis

14. What is the derived unit for force?

15. How do you ensure units cancel in analysis?

16. What does high precision indicate?

17. If a measurement is 3.456 and significant figures, what is the new value?

18. What is the slope formula?

19. Define interpolation.

20. In a bar graph, what does the y-axis represent?

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### Analyzing Data

Study Guide: Units and Measurements, Dimensional Analysis

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### Analyzing Data

Study Guide: Units and Measurements, Scientific Notation and Dimensional Analysis

**SI Measurements**  
The International System of Units (SI) is the modern form of the metric system and the most widely used system of measurement. It includes seven base units: meter (m) for length, kilogram (kg) for mass, second (s) for time, ampere (A) for electric current, kelvin (K) for temperature, mole (mol) for amount of substance, and candela (cd) for luminous intensity.

**SI Prefixes**  
SI prefixes are used to form decimal multiples and submultiples of SI units. Examples include:  
 • kilo- (k) =  $10^3$  = 1,000  
 • centi- (c) =  $10^{-2}$  = 0.01  
 • milli- (m) =  $10^{-3}$  = 0.001

**Converting from Celsius to Kelvin**  
To convert Celsius to Kelvin, add 273.15 to the Celsius temperature.  
 Example:  $25^\circ\text{C} + 273.15 = 298.15\text{ K}$

**Derived units are combinations of base units.**  
 Examples:  
 • Density =  $\frac{\text{mass}}{\text{volume}}$   
 • Speed =  $\frac{\text{distance}}{\text{time}}$   
 • Acceleration =  $\frac{\text{speed}}{\text{time}}$

**Density**  
Density is the mass per unit volume of a substance.  
 Density =  $\frac{\text{Mass}}{\text{Volume}}$   
 Common units are grams per cubic centimeter ( $\text{g/cm}^3$ ) or kilograms per cubic meter ( $\text{kg/m}^3$ ).

**Graphing**  
A bar graph is used to compare data across categories. The y-axis represents the quantity being measured, and the x-axis represents the categories.

# Study Guide Analyzing Data Chemistry Answer Key

**KJ Lindholm-Leary**



## **Study Guide Analyzing Data Chemistry Answer Key:**

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- Chapter 2: Essential Elements of Study Guide Analyzing Data Chemistry Answer Key
- Chapter 3: Study Guide Analyzing Data Chemistry Answer Key in Everyday Life
- Chapter 4: Study Guide Analyzing Data Chemistry Answer Key in Specific Contexts
- Chapter 5: Conclusion

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