

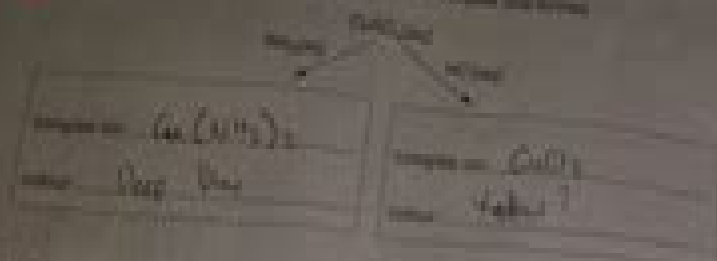
Question 1

Answer all the questions

The reaction is about the synthesis of copper complexes and ligands.

(a) The reaction shows the reaction of copper(II) sulfate with ligand.

In the boxes, write the formula and name of the complex and ligand.



(b) Cu^{2+} ions form a complex ion B with two ethylenediamine ligands and two water molecules. The ethylenediamine ion is a bidentate ligand.

The skeletal formula of the ethylenediamine ion is shown in Fig. 1.1 below.



Fig. 1.1

(i) What is meant by the term bidentate ligand?

A ligand with two donor atoms that can form two coordinate bonds with a central metal ion.

(ii) The complex ion A is shown in Fig. 1.2 below.

The structure of one of the ethylenediamine ligands is shown below. The charge has been given.



Complex A

Complete the 3D diagram of the other two geometries of A. You do not need to include any charges.

Indicate with ticks whether the stereoisomers are cis, trans, optical or a combination of these types.

Stereoisomer		
Type	cis trans optical	cis trans optical

(iii) What is the empirical formula, including the charge, of the complex ion A?

$[Cu(NH_3)_2(en)_2]^{2+}$

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JR Anderson



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