

Name _____ Date _____ Class _____

CHAPTER 4 STUDY GUIDE FOR CONTENT MASTERY**The Structure of the Atom****Section 4.1 Early Theories of Matter***In your textbook, read about the philosophers, John Dalton, and defining the atom.*For each statement below, write *true* or *false*.

- false** _____ 1. Ancient philosophers carefully performed controlled experiments.
- true** _____ 2. Philosophers formulated explanations about the nature of matter based on their own experiences.
- true** _____ 3. Both Democritus and Dalton suggested that matter is made up of atoms.
- true** _____ 4. Dalton's atomic theory stated that atoms separate, combine, or rearrange in chemical reactions.
- false** _____ 5. Dalton's atomic theory stated that matter is mostly empty space.
- false** _____ 6. Dalton was correct in thinking that atoms could not be divided into smaller particles.
- true** _____ 7. Dalton's atomic theory stated that atoms of different elements combine in simple whole-number ratios to form compounds.
- true** _____ 8. Dalton thought that all atoms of a specific element have the same mass.
- false** _____ 9. Democritus proposed that atoms are held together by chemical bonds, but no one believed him.
- true** _____ 10. Dalton's atomic theory was based on careful measurements and intensive research.
- false** _____ 11. There are no instruments powerful enough to supply atoms so that they can be seen.
- true** _____ 12. The smallest particle of an element that retains the properties of that element is called an atom.

Name _____ Date _____ Class _____

CHAPTER 4 STUDY GUIDE FOR CONTENT MASTERY**Section 4.2 Subatomic Particles and the Nuclear Atom***In your textbook, read about discovering the electron and the nuclear atom.*

For each item in Column A, write the letter of the matching item in Column B.

Column A	Column B
_____ 1. Proposed the nuclear atomic model	a. Thomson
_____ 2. Determined the mass-to-charge ratio of an electron	b. Millikan
_____ 3. Calculated the mass of an electron	c. Rutherford

Draw and label a diagram of each atomic model.

- a. plum pudding model**

Drawing should look like a ball of chocolate chip cookie dough. The chocolate chips should be labeled with negative charge or as electrons. The dough should be labeled as evenly distributed positive charges.

- b. nuclear atomic model**

Drawing should look like a peach with a pit. The pit should be labeled nucleus and should include labeled protons and neutrons. The outer circle of the peach should be labeled electrons.

In your textbook, read about the discovery of protons and neutrons.

Complete the following table of protons, electrons, and neutrons characteristics.

Particle	Symbol	Location	Relative Charge	Relative Mass
1. Proton	p^+	In the nucleus	$1+$	1
2. Neutron	n^0	In the nucleus	0	1
3. Electron	e^-	In the space surrounding the nucleus	$1-$	1/1840

Study Guide Answers Nuclear Change

Brian W. Stewart



Study Guide Answers Nuclear Change:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. Nuclear Cardiology Study Guide Andrzej Moniuszko, B. Adrian Kesala, 2013-10-19 This book presents a comprehensive review of nuclear cardiology principles and concepts necessary to pass the Nuclear Cardiology Technology Specialty Examination. The practice questions are similar in format and content to those found on the Nuclear Medicine Technology Certification Board (NMTCB) and American Registry of Radiological Technologists (ARRT) examinations, allowing test takers to maximize their chances of success. The book is organized by test sections of increasing difficulty, with over 600 multiple-choice questions covering all areas of nuclear cardiology, including radionuclides, instrumentation, radiation safety, patient care, and diagnostic and therapeutic procedures. Detailed answers and explanations to the practice questions follow. It also includes helpful test-taking tips. Supplementary appendices include commonly used abbreviations and symbols in nuclear medicine, a glossary of cardiology terms, and useful websites. Nuclear Cardiology Study Guide is a valuable reference for nuclear medicine technologists, nuclear medicine physicians, and all other imaging professionals in need of a concise review of nuclear cardiology. **Study Guide to Accompany Calculus for the Management, Life, and Social Sciences** Clyde Metz, 1984-01-01 Study Guide to Accompany Calculus for the Management, Life, and Social Sciences. *Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry* Katherine J. Denniston, Denniston Katherine, 2006. *Digital SAT Study Guide Premium, 2024: 4 Practice Tests + Comprehensive Review + Online Practice* Brian W. Stewart, 2023-10-03 Always study with the most up-to-date prep. Look for Digital SAT Study Guide Premium 2025 4 Practice Tests Comprehensive Review Online Practice ISBN 9781506292496 on sale July 2, 2024. Publisher's Note: Products purchased from third-party sellers are not guaranteed by the publisher for quality, authenticity, or access to any online entities included with the product. *Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e* Kenneth W. Raymond, 2005-10-07 Finally, readers have a shorter

less intimidating introduction to general organic and biological chemistry Not only is Raymond s text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry

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questions in that particular area which those taking the ABU examinations are likely to be tested on With a foreword

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