

# IONIC AND METALLIC BONDING

# Chapter Test A

#### A. Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

	Column A	Column B
E	1. electron dot structure	<ul> <li>a. a mixture of two or more elements, at least one of which is a metal</li> </ul>
F	2. ionic compound	<ul> <li>the number of ions of opposite charge surrounding each ion in a crystal</li> </ul>
H	3. valence electron	<ul> <li>the force of attraction binding oppositely charged ions together</li> </ul>
C	4. ionic bond	d. the attraction of valence electrons for positive metal ions
J	5. chemical formula	<ul> <li>a depiction of valence electrons around the symbol of an element</li> </ul>
		f. compound of cations and anions
G	6. halide ion	g. an anion of a halogen
Α	7. alloy	h. an electron in the highest occupied energy level of an atom
1	8. octet rule	<ol> <li>Atoms in most compounds tend to achieve the electron configuration of a noble gas.</li> </ol>
K	9. formula unit	
		<ol> <li>shows the kinds and numbers of atoms in the smallest representative unit of a substance</li> </ol>
В	10. coordination number	k. lowest whole-number ratio of ions in an ionic compound
D	11. metallic bond	ac sowest whose-manner ratio of some in an some compound

## **B. Multiple Choice**

Choose the best answer and write its letter on the line.

 1.2	. How man	y vadence	electrons	does an	atom of	amy e	element is	n Group 6.	A. harve?
	the state of the s					100			

a. 2 c. 6 b. 4 d. 8

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**Arshad Iqbal** 

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