

CHAPTER 8 STUDY GUIDE

Covalent Bonding

Section 8.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond

molecule

sigma bond

exothermic

pi bond

When sharing of electrons occurs, the attachment between atoms that results is called a(n) (1) _____. When such an attachment is formed, bond dissociation energy is released, and the process is (2) _____. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) (3) _____. If the electrons shared are centered between the two atoms, the attachment is called a(n) (4) _____. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) (5) _____.

In your textbook, read about single and multiple bonds and bond strength.

Circle the letter of the choice that best completes the statement or answers the question.

6. In what form do elements such as hydrogen, nitrogen, and oxygen normally occur?
 - a. as single atoms
 - b. as molecules containing two atoms
 - c. as molecules containing three atoms
 - d. as molecules containing four atoms
7. How many electrons are shared in a double covalent bond?
 - a. none
 - b. one
 - c. two
 - d. four
8. Bond length is the distance between
 - a. two molecules of the same substance.
 - b. the electrons in two attached atoms.
 - c. the nuclei of two attached atoms.
 - d. the orbitals of two attached atoms.
9. Which of the following relationships relating to bond length is generally correct?
 - a. the shorter the bond, the stronger the bond
 - b. the shorter the bond, the weaker the bond
 - c. the shorter the bond, the fewer the electrons in it
 - d. the shorter the bond, the lower the bond dissociation energy

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Table of Contents Study Guide Covalent Bonding Key

1. Understanding the eBook Study Guide Covalent Bonding Key
 - The Rise of Digital Reading Study Guide Covalent Bonding Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide Covalent Bonding Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide Covalent Bonding Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide Covalent Bonding Key
 - Personalized Recommendations
 - Study Guide Covalent Bonding Key User Reviews and Ratings
 - Study Guide Covalent Bonding Key and Bestseller Lists
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- 8. Staying Engaged with Study Guide Covalent Bonding Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide Covalent Bonding Key
- 9. Balancing eBooks and Physical Books Study Guide Covalent Bonding Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide Covalent Bonding Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Study Guide Covalent Bonding Key
 - Setting Reading Goals Study Guide Covalent Bonding Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Study Guide Covalent Bonding Key
 - Fact-Checking eBook Content of Study Guide Covalent Bonding Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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