

34	<p>a) How <math>^{35}\text{Cl}_{17}</math> and <math>^{37}\text{Cl}_{17}</math> are related to each other?</p> <p>b) If both of these forms of Cl occurs in nature as 75% and 25% respectively. Calculate the average atomic mass of Chlorine atom.</p> <p>c) Draw the labelled schematic atomic structure of Chlorine atom having <math>Z=17</math>.</p> <p style="text-align: center;">Or</p> <p>a) Distinguish between the properties of electrons, protons, and neutrons with respect to position, charge and mass.</p> <p>b) Write the valency and atomic symbol of the following element:</p> <p>i. Sodium (<math>Z=11</math>)</p> <p>ii. Aluminium (<math>Z=13</math>)</p> <p>iii. Potassium (<math>Z=19</math>)</p> <p>iv. Calcium (<math>Z=20</math>)</p>	5
35	<p>a) Plastids are present only in the plant. Which types of plastids are present in :</p> <p>i. Petals;            ii. Leaves;            iii. fruit pulp</p> <p>b) What happen if chloroplast is removed from the plant. Explain.</p> <p style="text-align: center;">Or</p> <p>To test the effect of the concentration of the solution on the cell a student placed some plant cell and animal cell (RBC) for 15 minute in the hypertonic solution and then shifted them into hypotonic solution.</p> <p>Answer the following question:</p> <p>a) What is hypotonic solution.</p> <p>b) What will the expected observation in the above situation.</p> <p>c) what is the role of cell wall in the cell in above situation.</p>	5
36	<p>a) Explain universal law of gravitation also write its mathematical expression.</p> <p>b) Write three importance of this law.</p> <p style="text-align: center;">Or</p> <p>Give reason:</p> <p>a) carry a bag with broader strap is easy than a thin strap bag.</p> <p>b) why sharp knife cut easy in comparison with the blunt knife.</p> <p>c) camel can run in a desert easily.</p>	5
<p><b>SECTION-E</b></p> <p>Q.no.37 to 39 are case based / source- based questions with 2 to 3 short sub - parts.</p>		

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