

Name: \_\_\_\_\_

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## Classifying Chemical Reactions Lab

**Educational Goals:** Students will understand...

- How to distinguish between a single displacement, double displacement, composition and decomposition reaction
- How to observe reactions and predict the product

**Student Objectives:** Students will...

- Perform a composition, decomposition, single displacement and double displacement reaction
- Make observations of chemical reactions and categorize them
- Write and balance chemical equations

### Pre-lab Questions

For each of the following experimental procedures and observations:

- a) Write a balanced chemical equation
- b) Name the product
- c) Identify the type of reaction

1. A student heats barium metal over a flame, and begins to react with oxygen gas in the air. A white, crystalline solid begins to form on the barium.

2. A chemist mixes lead (II) nitrate and potassium iodide solutions (both of which are clear liquids) and mixes them. A yellow precipitate appears in the final solution. The remaining liquid is an ionic solution. Solid precipitates containing lead are always yellow.

3. Magnesium metal is placed in sulfuric acid. The solution begins to bubble. The remaining liquid is an ionic solution.

# Pre Lab Answers To Classifying Chemical Reactions

**Sabine Zange**



## **Pre Lab Answers To Classifying Chemical Reactions:**

**Exploring Physical Science in the Laboratory** John T. Salinas ,2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging Exploring Physical Science in the Laboratory guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts

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,1986 **Safety-Scale Lab Exp Biochem 2e** Spencer L. Seager,Michael R. Slabaugh,1994-05 *Prentice Hall Physical*

*Science Concepts in Action Program Planner National Chemistry Physics Earth Science* ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

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Changes in the Junior High Science Classroom Keith Alan Morris, 1998      *U.S. Government Research & Development Reports*, 1967      **Electroanalytical Abstracts**, 1977 International journal dealing with the documentation of all aspects of fundamental physico chemical and analytical electrochemistry      **Selected Water Resources Abstracts**, 1989      **U.S. Government Research Reports**, 1962      **Process Quality Control** Ellis R. Ott, Edward G. Schilling, Dean V. Neubauer, 2005-02-08 Ellis Ott taught generations of quality practitioners to be explorers of the truth through the collection and graphical portrayal of data From a simple plea to plot the data to devising a graphical analytical tool called the analysis of means ANOM Ott demonstrated that process knowledge is to be gained by seeking the information contained within the data In this newest version of Ott's classic text the authors have strived to continue down the path that he created for others to follow Additions to this revised edition include the use of dot plots as an alternative to histograms digidot plots adding events to charts emphasis on the role that acceptance control charts play in controlling risks and the computation of average run length ARL a new chapter devoted to process capability process performance and process improvement including the use of confidence intervals for process capability metrics narrow limit gauging as another means of assessing the capability of a process Six Sigma methodology design resolution scatter plot matrices as applied to datasets of higher dimensions and a new chapter on measurement studies

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