

- B. less volume
- C. same volume
- D. unpredictable behaviour
- E. no such correlation.

**Answer: A**

**10. General gas equation is**

- A.  $PV=nRT$
- B.  $PV=mRT$
- D.  $PV = C$
- C.  $PV=KRT$
- E.  $C_p-C_v = W_j$

**Answer: B**

**11. According to Dalton's law, the total pressure of the mixture of gases is equal to**

- A. greater of the partial pressures of all
- B. average of the partial pressures of all
- C. sum of the partial pressures of all
- D. sum of the partial pressures of all divided by average molecular weight
- E. atmospheric pressure.

**Answer: C**

**12. Which of the following can be regarded as gas so that gas laws could be applicable, within the commonly encountered temperature limits.**

- A.  $O_2$ ,  $N_2$ , steam,  $CO_2$
- B.  $O_2$ ,  $N_2$ , water vapour
- C.  $SO_2$ ,  $NH_3$ ,  $CO_2$ , moisture
- D.  $O_2$ ,  $N_2$ ,  $H_2$ , air
- E. steam vapours,  $H_2$ ,  $CO_2$ .

**Answer: D**

**13. The unit of pressure in S.I. units is**

- A.  $kg/cm^2$
- B. mm of water column
- C. pascal
- D. dynes per square cm
- E. bars

**Answer: C**

**14. A closed system is one in which**

- A. mass does not cross boundaries of the system, though energy may do so
- B. mass crosses the boundary but not the energy
- C. neither mass nor energy crosses the boundaries of the system
- D. both energy and mass cross the boundaries of the system
- E. thermodynamic reactions take place.

**Answer: A**

**15. Temperature of a gas is produced due to**

- A. its heating value
- B. kinetic energy of molecules

# Thermodynamics Multiple Choice Questions And Answers

**R. Yadav**



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