Types of Chemical Reaction Worksheet

Name:

Balance the reactions and indicate which types of chemical reaction are being represented:

a) ____ NaBr + ____ Ca(OH)₂ → ___ CaBr₂ + ____ NaOH Reaction Type : _____

b) ____ NH₃+ ____ H₂SO₄ → ____ (NH₄)₂SO₄ Reaction Type : ______

c) ____C₉H₉O + ____O₂ → ____CO₂ + ____H₉O Reaction Type : ______

d) ____ Pb + ___ H₃PO₄ → ___ H₂ + ___ Pb₃(PO₄)₂ Reaction Type : ______

e) ____ Li₃N + ___ NH₄NO₃ → ___ LiNO₃ + ___ (NH₄)₂N Reaction Type : _____

f) HBr + Al(OH)₃ → H₂O + AlBr₃ Reaction Type :

g) Na₃PO₄ + KOH -> NaOH + K₃PO₄ Reaction Type

h) ___ MgCl₂ + ___Li₂CO₃ → ___ MgCO₃ + ___ LiCl Reaction Type ______

i) C₆H₁₂ + O₂ → CO₂ + H₂O Reaction Type

i) Pb + FeSO₄ → PbSO₄ + Fe Reaction Type _____

k) CaCO₃ → CaO + CO₂ Reaction Type _____

m) RbNO₃ + BeF₂ → Be(NO₃)₂ + RbF Reaction Type _____

n) AgNO₃ + Cu → Cu(NO₃)₂ + Ag Reaction Type_____

o) C₃H₆O + O₂ → CO₂ + H₂O Reaction Type ______

p) C₅H₅ + Fe → Fe(C₅H₅)₂ Reaction Type

q) SeCl₆ + O₂ → SeO₂ + Cl₂ Reaction Type ______

r) Mgl₂ + Mn(SO₃)₂ → MgSO₃ + Mnl₄ Reaction Type _____

s) O₃ → O + O₂ Reaction Type

t) NO₂ → O₂ + N₂ Reaction Type _____

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