

**Syllabus reference:**

*"Write ionic half-equations and use them to construct full ionic equations."*

**Aim of this worksheet:**

To introduce you to the idea of balancing redox equations by adjusting coefficients and by adding  $\text{OH}^-$ ,  $\text{H}_2\text{O}$ , and  $\text{H}^+$  where necessary

**Task: Balance the following redox reactions:**

- $\text{SO}_2 (\text{g}) + \text{HNO}_2 (\text{aq}) \rightarrow \text{H}_2\text{SO}_4 (\text{aq}) + \text{NO} (\text{g})$
- $\text{Al} (\text{s}) + \text{H}_2\text{SO}_4 (\text{aq}) \rightarrow \text{Al}_2(\text{SO}_4)_3 (\text{aq}) + \text{H}_2 (\text{g})$
- $\text{Au}^{3+} (\text{aq}) + \text{I}^- (\text{aq}) \rightarrow \text{Au} (\text{s}) + \text{I}_2 (\text{s})$
- $\text{S}^{2-} (\text{aq}) + \text{I}_2 (\text{s}) \rightarrow \text{SO}_4^{2-} (\text{aq}) + \text{I}^- (\text{aq})$
- $\text{Br}_2 (\text{aq}) + \text{OH}^- (\text{aq}) \rightarrow \text{Br}^- (\text{aq}) + \text{BrO}_3^- (\text{aq})$
- $\text{H}_2\text{O}_2 (\text{aq}) + \text{ClO}_4^- (\text{aq}) \rightarrow \text{O}_2 (\text{g}) + \text{ClO}_2^- (\text{aq})$
- $\text{Mn} (\text{s}) + \text{HNO}_3 (\text{aq}) \rightarrow \text{Mn}^{2+} (\text{aq}) + \text{NO}_2 (\text{g})$
- $\text{I}_2 (\text{s}) + \text{OCl}^- (\text{aq}) \rightarrow \text{IO}_3^- (\text{aq}) + \text{Cl}^- (\text{aq})$
- $\text{Cr}_2\text{O}_7^{2-} (\text{aq}) + \text{HNO}_2 (\text{aq}) \rightarrow \text{Cr}^{3+} (\text{aq}) + \text{NO}_3^- (\text{aq})$
- $\text{CrO}_4^{2-} (\text{aq}) + \text{S}^{2-} (\text{aq}) \rightarrow \text{Cr}(\text{OH})_3 (\text{s}) + \text{S} (\text{s})$

# Redox Equations Worksheet With Answers

**NARAYAN CHANGDER**



## **Redox Equations Worksheet With Answers:**

**Revise A2 Chemistry for AQA** Paddy Gannon, 2005-04 Part of the series of AS and A2 revision guides this title gives students what they need to know for the AQA exams It includes material organised into bite sized chunks of information

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*Reaction-kinetic Studies of Inorganic Redox Reactions in Solution* S. W. Gilks, 1960      [Basic Principles of Inorganic Redox Reactions in Aqueous Solutions](#) A.S. Ghosh Mazumdar, 1971      **Thermal Redox Reactions Between Metal Ions and Radicals in Aqueous Solutions** Zuhair Kamal Jamil, 1965      [Chemistry Equations And Answers \(Speedy Study Guides\)](#) Speedy Publishing, 2015-04-24

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