

Question 1: (10 marks)

- 1.1 Name three methods to control corrosion. (3)
- 1.2 Briefly explain and complete the following Table 1 in the answer sheet. (7)

Corrosion	Corrosion
Electrochemical	Electrochemical
Galvanic	Galvanic
Atmospheric	Atmospheric
Microbiological	Microbiological
Stress Corrosion	Stress Corrosion

(3+4) (3)

Question 2: (10 marks)

Describe the following types of corrosion and give a practical example of each.

- 2.1 Pitting corrosion (3)
- 2.2 Pitting corrosion (3)

Question 3: (10 marks)

- 3.1 Name three corrosion inhibitors to keep metals and alloys away from metals which are active. (3)

- 3.2 Name the chemical composition of the following compounds:

3.2.1 Corrosion inhibitor

3.2.2 Corrosion inhibitor

(2+4) (3)

- 3.3 Name a material that is a good conductor of electricity. (1)

- 3.4 Name three properties of water. (3)

Question 4: (10 marks)

- 4.1 Use pH to explain corrosion. (3+4) (3)
- 4.2 Give three examples of reactions that cause corrosion. (3+4) (3)

Question 5: (10 marks)

- 5.1 Name a metal, identify its use and describe the function of its alloy. (3)

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