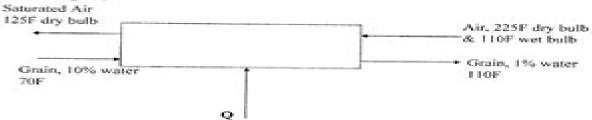
- Air at 45°C and 10% relative humidity is to be humidified adiabatically to 60% relative humidity. Use the psychometric chart to obtain the following numbers:
 - (a) (5 points) The dry bulb and adiabatic saturation temperature of the final air.
 - (b) (5 points) The dew point of the final air
 - (c) (5 points) The absolute humidity and humid volume of final air.
 - (d) (5 points) The rate at which water must be added to humidify 15m3/min of the entering air.
- 2. (20 points) Two aqueous sulfuric acid solutions containing 20.0 wt% H2SO4 (specific gravity=1.139) and 60.0 wt % H₂SO₄ (specific gravity=1.498) are mixed to form a 4.00 molar solution (specific gravity=1.213). Assuming 100 kg of 20% solution as the basis calculate:
 - (a) (10 points) the mass fraction of sulfuric acid in the product solution.
 - (b) (10 points) the feed ratio (liters of 20% solution/liter of 60% solution).
- 3. (30 points) A rotary dryer operating at atmospheric pressure dries 10 tons/day of wet grain at 70F, from a moisture content of 10% to 1% moisture. The air flow is countercurrent to the flow of grain, enters at 225F dry bulb and 110F wet-bulb temperature, and leaves at 125F dry-bulb (see figure).



Determine: (a) the humidity of entering and leaving air if the latter is saturated, (b) the water removal in pounds per hour. (c) the daily product output in pounds per day, and (d) the heat input to the dryer. Assume that there is no heat loss from the dryer, that the grain is discharged at 110F, and that its specific heat is 0.18.

- 4. (30 points) A fuel gas containing 85 mole % methane and the balance ethane is burned completely with pure oxygen at 25 °C at latm. Suppose the reactor is continuous. Take a basis of calculation of I mol/s of the fuel gas, combusted with stoichiometric amount of oxygen. (a) (10 points) Please calculate the rate at which heat must be transferred from the reactor if the
 - products are brought back down to 25 °C.
 - (b) (15 points) Please calculate the final temperature of the combustion gas if the process is under adiabatic condition at constant pressure 1atm.
 - (c) (5 points) Please calculate the final volume of the combustion gas if the process is under adiabatic condition at constant pressure latm. If the volume of the reactor can only be constant and the pressure can change, what do you expect to happen to the reactor?

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Christine Tina Chasek, Thomas Z. Maxson

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Urden, Kathleen M. Stacy, Mary E. Lough, 2021-02-18 Winner of the 2nd issues place American Journal of Nursing Book of the Year award in emergency critical care nursing for 2021 Selected for Doody's Core Titles 2024 with Essential Purchase designation in Critical Care Prepare for success in today s high acuity progressive and critical care settings Critical Care Nursing Diagnosis and Management 9th Edition helps you understand and apply critical care nursing principles and concepts to clinical assessment diagnostic procedures and therapeutic management Known for its comprehensive coverage this leading textbook uses a logical body systems organization to address the care of patients with physiological alterations New to this edition are illustrated Patient Centered Critical Care features aimed at humanizing the ICU as well as Next Generation NCLEX Exam style case studies to help you further develop your clinical judgment skills and prepare for the latest nursing licensure exam Also ideal for CCRN and PCCN exam preparation this book is a one stop resource on the concepts and skills required for critical care nursing Time tested high quality content addresses all aspects of today s high acuity progressive and critical care nursing Consistent organization within each body system unit provides an efficient framework for learning for CCRN and PCCN certification preparation and for reference in clinical practice Comprehensive evidence based content is highly referenced and includes internet resources for further research and study Enhanced Quality and Safety Education for Nurses QSEN integration links text content to QSEN competencies through the addition of QSEN related questions in case studies QSEN labeled features and boxes QSEN content icons and highlighted QSEN information

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