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# **Pacific Senior Secondary Certificate** CHEMISTRY

## 2011

## OUESTION AND ANSWER BOOKLET

Time allowed: 3 hours

This Examination Paper consists of TWO sections, ATTEMPT ALL QUESTIONS.

		MARKS	THME
SECTION A	Multiple Choice Questions	40	36 mins
SECTION B	Q21: Atomic Structure and Bonding	2.5	22 mins
	Q22: Quantitative Chemistry	28	24 mins
	Q23: Organic Chemistry	20	18 mins
	Q24: More Organic Chemistry	20	18 mins
	Q25: Inorganic Chemistry	21	20 mins
	Q26: Oxidation and Reduction	2.5	22 mins
	Q27: Principles of Physical Chemistry	21	20 mins
PROPERTY ACTOR		200	190 malore

- 2. Write your Student Personal Identification Number (SPIN) on the top right hand corner of this page and at the top of the fold-out flap on the last page.
- Write all answers to the Multiple Choice questions on the answer sheet on the FOLD-OUT 3. FLAP on the last page.
- 4. In SECTION B, write the answers to the questions in the spaces provided in this booklet.
  - A copy of the Periodic Table of the Elements Sheet No. 3/2 is provided.
  - NOTE: The symbol M is used for molar mass.
  - For example,  $M(Na) = 23 \text{ g mol}^{-1}$  and  $M(CO_2) = 44 \text{ g mol}^{-1}$

Check that this booklet contains pages 2-29 in the correct order and none of these pages is blank. Page 30 has been left blank deliberately.

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