

## CHEMISTRY FINAL EXAM REVIEW GUIDE 2012 Scott

**DISCLAIMER: THIS IS A SAMPLE OF QUESTIONS/PROBLEMS FROM THIS YEAR. PLEASE BE SURE TO REVIEW YOUR NOTES, LABS, COJ'S, OBJECTIVE SHEETS, ETC. DO NOT RELY ONLY ON THIS REVIEW GUIDE.**

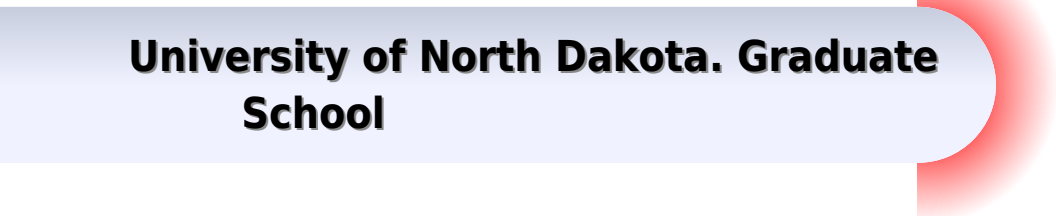
### Multiple Choice

*Identify the choice that best completes the statement or answers the question.*

- \_\_\_\_\_ 1) Which of the following observations is quantitative?  
a. The liquid turns blue litmus paper red.  
b. The liquid boils at 100°C.  
c. The liquid tastes bitter.  
d. The liquid is cloudy.
- \_\_\_\_\_ 2) The symbols for units of length in order from largest to smallest are  
a. m, cm, mm, km. c. km, mm, cm, m.  
b. mm, m, cm, km. d. km, m, cm, mm.
- \_\_\_\_\_ 3) A volume of 1 milliliter is equivalent to  
a. 1 cubic centimeter. c. 1 liter.  
b. 1 gram. d.  $10^{-3}$  cubic decimeters.
- \_\_\_\_\_ 4) Which pair of quantities determines the density of a material?  
a. mass and weight c. volume and concentration  
b. volume and weight d. volume and mass
- \_\_\_\_\_ 5) A sample of gold has a mass of 96.5 g and a volume of 5.00 cm<sup>3</sup>. The density of gold is  
a. 0.0518 g/cm<sup>3</sup>. c. 101.5 g/cm<sup>3</sup>.  
b. 19.3 g/cm<sup>3</sup>. d. 483 g/cm<sup>3</sup>.
- \_\_\_\_\_ 6) The number of grams equal to 0.5 kg is  
a. 0.0005. c. 500.  
b. 0.005. d. 5000.
- \_\_\_\_\_ 7) If 1 inch equals 2.54 cm, how many centimeters equal 1 yard?  
a. 0.0706 cm c. 30.5 cm  
b. 14.2 cm d. 91.4 cm
- \_\_\_\_\_ 8) A chemist obtained the values 5.2246 g, 5.2353 g, and 5.2501 g for the mass of a sample. Without knowing the true mass of the sample, it can be said that these values have  
a. good precision.  
b. good accuracy.  
c. poor precision.  
d. poor accuracy.
- \_\_\_\_\_ 9) The measurement 0.035550 g rounded off to two significant figures would be  
a. 0.03 g. c. 0.036 g.  
b. 0.35 g. d.  $3.5 \times 10^2$  g.
- \_\_\_\_\_ 10) When 64.4 is divided by 2.00, the correct number of significant figures in the result is

# Scope For Chemistry Final Examination

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**Federal Register** ,2013-04     *Cumulative Supplement to the Code of Federal Regulations of the United States of America* ,1943     **The Code of Federal Regulations of the United States of America Having General Applicability and Legal Effect in Force June 1, 1938** ,1939     **General Orders** United States. War Department,1910     **Supplement to the Code of Federal Regulations of the United States of America** ,1939     **University of Michigan Official Publication** University of Michigan,1973 Each number is the catalogue of a specific school or college of the University

*Special Regulations* United States. War Department,1917     *Traceability in Chemical Measurement* Paul De Bièvre,Helmut Günzler,2005-12-06 Metrological traceability of chemical measurement results means the establishment of a relation to metrological stated references through an unbroken chain of comparisons This volume collects 56 outstanding papers on the topic mostly published in the period 2000 2003 in the journal Accreditation and Quality Assurance They provide the latest understanding and possibly the rationale why it is important to integrate the concept of metrological traceability including suitable measurement standards such as certified reference materials into the standard measurement procedures of every analytical laboratory In addition this anthology considers the benefits to both the analytical laboratory and the user of the measurement results

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