

Chemistry Spring Final Study Guide

Unit 10: Acids & Bases

Use the following information and complete the structure of Acids & Bases.

Acids	Bases
<p>Unit 10: Acids & Bases</p> <p>Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below.</p> <p>Label the y-axis.</p> <p>Label the x-axis.</p> <p>1. A solution of the perfect...</p> <p>2. A solution of...</p> <p>3. A solution of...</p> <p>4. The gas is...</p> <p>5. The solution...</p>	<p>1. Draw a line... (text partially obscured)</p> <p>2. Draw a line... (text partially obscured)</p> <p>3. Draw a line... (text partially obscured)</p> <p>4. Draw a line... (text partially obscured)</p> <p>5. Draw a line... (text partially obscured)</p>

Chemistry Spring Final Study Guide

Name: _____

Unit 1: Measurement Analysis

Mark the best answer from those provided to each question.

Mark the best answer from those provided to each question.

Determine the number of Significant Figures in each of the following numbers:

Mass = 1. _____ kg	Mass = 1.00 _____ kg
Volume = 1.0000 _____ L	Volume = 1.0000 _____ L

Convert the following into Scientific Notation:

Mass = 1. _____ kg	Mass = 1.00 _____ kg
Volume = 1.0000 _____ L	Volume = 1.0000 _____ L

Measurement Analysis Mark your answers to 3 places after the decimal. Show work to show your work.

1. Convert the 1000 meters into feet.

2. Convert 1000 meters into feet.

Unit 2: Reaction

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

H_2SO_4

Atomic Masses:

H=1.0
S=32.1
O=16.0

Reaction

Atomic Masses:

H=1.0
S=32.1
O=16.0

Write the empirical formula for each compound.

Compound	Empirical
C_2H_6	
C_4H_{10}	
C_6H_{12}	
C_8H_{18}	

Write the chemical formula for the compound.

Diagram showing the relationship between the compound and its elements:

```

graph TD
    A[Compound] --> B[Element]
    A --> C[Element]
    A --> D[Element]
  
```

Spring Final Chemistry Study Guide Answer Key

Frank Lees



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