## **Practice Final Exam Chem 112**

#### Multiple Choice

Identify the choice that best completes the statement or answers the question.

 Three hundred milliliters (150 mL) of 4.50 M sugar solution is diluted to 600 mL by the addition of pure water. Which of these statements about the solution is correct?

a. The moles of water remain the same.
 b. The moles of sugar decrease.
 c. The concentration of sugar increases to 5.0 M.
 d. The moles of sugar remain constant.
 e. All of these statements are incorrect.

How many moles of solute are present in 75.0 grams of solution that is 1.30% by mass glucose (C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>)?
 MM(C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>) = 180.16 g/mol
 a, 5.4 × 10<sup>-3</sup> moles b, 0.800 moles c, 176 moles d, 1.76 × 10<sup>-3</sup> moles e, 225 moles

A concentrated reagent is sold as a solution that is 60.0 mass % solute. If the molar mass of the solute is 244 g/mol and the density of this solution is 1.20 g/mL, what is the molarity of the solution?
 a. 10.0 M
 b. 2.95 M
 c. 15.5 M
 d. 0.72 M
 e. 1.45 M

- 4. The boiling point of pure benzene (C<sub>6</sub>H<sub>6</sub>) is 80.10 °C. A solution of 14.2 g of a solid dissolved in 64.0 g of benzene boils at 81.09 °C. If k<sub>6</sub> for benzene is 2.53 °C/molal, the molar mass of the solid is: a, 662 g/mol b, 56.0 g/mol c, 92.0 g/mol d, 628 g/mol e, 568 g/mol
- 5. What is the freezing point for a 0.71 m solution of glycerol (C<sub>3</sub>H<sub>6</sub>O<sub>5</sub>) in ethanol? Freezing point(ethanol) = -114.6 °C; K<sub>6</sub>(ethanol) = 1.99 °C/m
  a. 0.10 °C
  b. -113.2 °C
  c. -1.41 °C
  d. -116.0 °C
  e. none of these
- 6. At 450 °C, K<sub>c</sub> = 0.231 for the reaction:

 $2 \text{ SO}_2(g) + O_2(g) \rightleftharpoons 2 \text{ SO}_1(g)$ 

What is the value of the equilibrium constant for the following reaction at 450 °C?

$$SO_2(g) + \frac{1}{2}O_2(g) \rightleftharpoons SO_2(g)$$

$$K_c = ?$$

a. 0.481 b. 0.156 c. 0.231 d. 0.053 e. 4.329

Consider the following system at equilibrium and its corresponding concentration equilibrium constant:

$$3 O_2(g) \rightleftarrows 2 O_3(g)$$

$$K_c = 1.8 \times 10^{-6} \text{ at } 470 \, ^{\circ}\text{C}$$

What is the value of  $K_p$  for this reaction at 470 °C? a.  $K_p = 3.80 \times 10^6$  b.  $K_p = 8.53 \times 10^6$  c.  $K_p = 4.32 \times 10^6$  d.  $K_p = 1.10 \times 10^4$  e.  $K_p = 1.19 \times 10^4$ 

For the reaction l<sub>2</sub> (g) + Cl<sub>2</sub> (g) ₹<sup>2</sup> 2 ICl (g), the relation between K<sub>p</sub> and K<sub>r</sub> is:
 a, K<sub>n</sub> = K<sub>r</sub>
 b, K<sub>n</sub> = K<sub>r</sub>(RT)<sup>2</sup>
 c, K<sub>n</sub> = K<sub>r</sub>(RT)<sup>2</sup>
 d, K<sub>r</sub> = K<sub>r</sub>(RT)<sup>n</sup>
 e, K<sub>n</sub> = K<sub>r</sub>(RT)<sup>n</sup>

# **Spring 2013 Chemistry Final Exam Answers**

H. Niki, K.H. Becker

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troposphere take place Advances in Clinical Chemistry, 2025-04-05 Advances in Clinical Chemistry Volume 126 the latest installment in this internationally acclaimed series contains chapters authored by world renowned clinical laboratory scientists physicians and research scientists Chapters in this new release include Internalization of Extracellular Tau Oligomers in Alzheimer's Disease Advances in Neurofilament Light Chain Analysis Biomarkers of Endometriosis Lupus Nephritis II Treatment and Monitoring and Emerging Biomarkers in Type 2 Diabetes Mellitus Coagulopathy Prediction in Traumatic Brain Injury Provides the most up to date technologies in clinical chemistry and clinical laboratory science Authored by world renowned clinical laboratory scientists physicians and research scientists Presents the international benchmark for novel analytical approaches in the clinical laboratory **Chemistry Education and Sustainability in the** Global Age Mei-Hung Chiu, Hsiao-Lin Tuan, Hsin-Kai Wu, Jing-Wen Lin, Chin-Cheng Chou, 2012-12-05 This edited volume of papers from the twenty first International Conference on Chemical Education attests to our rapidly changing understanding of the chemistry itself as well as to the potentially enormous material changes in how it might be taught in the future Covering the full range of appropriate topics the book features work exploring themes as various as e learning and innovations in instruction and micro scale lab chemistry In sum the 29 articles published in these pages focus the reader s attention on ways to raise the quality of chemistry teaching and learning promoting the public understanding of chemistry deploying innovative technology in pedagogy practice and research and the value of chemistry as a tool for highlighting sustainability issues in the global community Thus the ambitious dual aim achieved in these pages is on the one hand to foster improvements in the leaching and communication of chemistry whether to students or the public and secondly to promote advances in our broader understanding of the subject that will have positive knock on effects on the world's citizens and environment In doing so the book addresses as did the conference the neglect suffered in the chemistry classroom by issues connected to globalization even as it outlines ways to bring the subject alive in the classroom through the use of innovative technologies Beer in Maryland Maureen O'Prey, 2018-03-05 This history begins with the earliest brewers in the colony women revealing details of the Old Line State's brewing families and their methods Stories never before told trace the effects of war competition the Industrial Revolution Prohibition and changing political philosophies on the brewing industry Some brewers persevered through crime scandal and intrigue to play key roles in building their communities Today s craft brewers face a number of very different challenges from monopolistic macro breweries and trademark quandaries to hop shortages while attempting to establish their own legacies **The Multiplayer Classroom** Lee Sheldon, 2020-03-09 Go beyond gamification s badges and leaderboards with the new edition of the book first published in 2011 that helped transform education Going far beyond the first edition of The Multiplayer Classroom forthrightly examining what worked and what didn t over years of development here are the tools to design any structured learning experience as a game to engage your students raise their grades and ensure their attendance Suitable for use in the classroom or the boardroom this book

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