

Periodic Table of the Elements

1 IA H Hydrogen (1.008)																	18 VIIIA He Helium (4.003)		
3 IIA Li Lithium (6.941)	4 IIA Be Beryllium (9.012)													5 IIIA B Boron (10.81)	6 IIIA C Carbon (12.011)	7 IVA N Nitrogen (14.007)	8 IVA O Oxygen (15.999)	9 VA F Fluorine (18.998)	10 VIA Ne Neon (20.180)
11 IA Na Sodium (22.990)	12 IIA Mg Magnesium (24.305)													13 IIIA Al Aluminum (26.982)	14 IIIA Si Silicon (28.086)	15 IVA P Phosphorus (30.974)	16 IVA S Sulfur (32.06)	17 VA Cl Chlorine (35.45)	18 VIA Ar Argon (39.948)
19 IA K Potassium (39.098)	20 IIA Ca Calcium (40.078)	21 IIB Sc Scandium (44.956)	22 IIIB Ti Titanium (47.88)	23 IIIB V Vanadium (50.942)	24 IIIB Cr Chromium (51.996)	25 IIIB Mn Manganese (54.938)	26 IIIB Fe Iron (55.845)	27 IIIB Co Cobalt (58.933)	28 IIIB Ni Nickel (58.693)	29 IIB Cu Copper (63.546)	30 IIB Zn Zinc (65.38)	31 IIB Ga Gallium (69.723)	32 IIB Ge Germanium (72.63)	33 IIIA As Arsenic (74.922)	34 IIIA Se Selenium (78.96)	35 IIIA Br Bromine (79.904)	36 IIIA Kr Krypton (83.798)		
37 IA Rb Rubidium (85.468)	38 IIA Sr Strontium (87.62)	39 IIB Y Yttrium (88.906)	40 IIIB Zr Zirconium (91.224)	41 IIIB Nb Niobium (92.906)	42 IIIB Mo Molybdenum (95.94)	43 IIIB Tc Technetium (98.906)	44 IIIB Ru Ruthenium (101.07)	45 IIIB Rh Rhodium (102.91)	46 IIIB Pd Palladium (106.91)	47 IIB Ag Silver (107.87)	48 IIB Cd Cadmium (112.41)	49 IIB In Indium (114.82)	50 IIB Sn Tin (118.71)	51 IIIA Sb Antimony (121.76)	52 IIIA Te Tellurium (127.6)	53 IIIA I Iodine (126.91)	54 IIIA Xe Xenon (131.29)		
55 IA Cs Cesium (132.91)	56 IIA Ba Barium (137.33)	57-71 Lanthanides	72 IIIB Hf Hafnium (178.49)	73 IIIB Ta Tantalum (180.95)	74 IIIB W Tungsten (183.84)	75 IIIB Re Rhenium (186.21)	76 IIIB Os Osmium (190.23)	77 IIIB Ir Iridium (192.22)	78 IIIB Pt Platinum (195.08)	79 IIB Au Gold (196.97)	80 IIB Hg Mercury (200.59)	81 IIB Tl Thallium (204.38)	82 IIB Pb Lead (207.2)	83 IIIA Bi Bismuth (208.98)	84 IIIA Po Polonium (209)	85 IIIA At Astatine (210)	86 IIIA Rn Radon (222)		
87 IA Fr Francium (223)	88 IIA Ra Radium (226)	89-103 Actinides	104 IIIB Rf Rutherfordium (261)	105 IIIB Db Dubnium (262)	106 IIIB Sg Seaborgium (266)	107 IIIB Bh Bohrium (264)	108 IIIB Hs Hassium (277)	109 IIIB Mt Meitnerium (268)	110 IIIB Ds Darmstadtium (271)	111 IIB Rg Roentgenium (272)	112 IIB Cn Copernicium (285)	113 IIB Nh Nihonium (284)	114 IIB Fl Flerovium (289)	115 IIIA Mc Moscovium (288)	116 IIIA Lv Livermorium (293)	117 IIIA Ts Tennessine (289)	118 IIIA Og Oganesson (294)		

Atomic Number →

→ Symbol

→ Name

→ Atomic Weight

→ Electrons per shell

1

H

Hydrogen

1.008

1

57 La Lanthanum (138.91)	58 Ce Cerium (140.12)	59 Pr Praseodymium (140.91)	60 Nd Neodymium (144.24)	61 Pm Promethium (145)	62 Sm Samarium (150.36)	63 Eu Europium (151.96)	64 Gd Gadolinium (157.25)	65 Tb Terbium (158.93)	66 Dy Dysprosium (162.50)	67 Ho Holmium (164.93)	68 Er Erbium (167.26)	69 Tm Thulium (168.93)	70 Yb Ytterbium (173.05)	71 Lu Lutetium (174.97)
89 Ac Actinium (227)	90 Th Thorium (232.04)	91 Pa Protactinium (231.04)	92 U Uranium (238.03)	93 Np Neptunium (237.05)	94 Pu Plutonium (244.06)	95 Am Americium (243.06)	96 Cm Curium (247.07)	97 Bk Berkelium (247.07)	98 Cf Californium (251.08)	99 Es Einsteinium (252.08)	100 Fm Fermium (257.10)	101 Md Mendelevium (258.11)	102 No Nobelium (259.10)	103 Lr Lawrencium (262.11)

Section 3 The Periodic Table

Leo J. Malone, Theodore Dolter



Section 3 The Periodic Table:

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dynamic sphere of knowledge supporting a landscape of constantly changing hues Experimental boundaries are being shifted with exciting reductions in uncertainty at a staggering pace This text develops the Electro Gravi Magnetic EGM construct to define relationships between the distributions of mass energy over space time of fundamental particles The correlation of EGM calculations for mass size to experimental evidence is astonishing to at least four orders of magnitude greater than can be physically measured Most of the contents herein have been peer reviewed published in scientific literature For particle enthusiasts this text is a must **Methods of Soil Analysis, Part 3** D. L. Sparks, A. L. Page, P. A. Helmke, Richard H. Loeppert, 2020-01-22 A thorough presentation of analytical methods for characterizing soil chemical properties and processes

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