

The color of the chemical symbol indicates the physical state at room temperature. Carbon is a solid.

6 — Atomic number

**C** — Chemical symbol

Carbon — Element name

12.0 — Atomic mass

The background color indicates the type of element. Carbon is a nonmetal.

Period 1	<div>1 <b>H</b> Hydrogen 1.0</div>		type of element. Carbon is a nonmetal.																<div>18 <b>He</b> Helium 4.0</div>		Group 18													
	Group 1	Group 2	<div>Background</div> <div>Metals<div></div></div> <div>Metalloids<div></div></div> <div>Nonmetals<div></div></div>										<div>Chemical symbol</div> <div>Solid<div></div></div> <div>Liquid<div></div></div> <div>Gas<div></div></div>																Group 13	Group 14	Group 15	Group 16	Group 17	
Period 2	<div>3 <b>Li</b> Lithium 6.9</div>	<div>4 <b>Be</b> Beryllium 9.0</div>																	<div>5 <b>B</b> Boron 10.8</div>	<div>6 <b>C</b> Carbon 12.0</div>	<div>7 <b>N</b> Nitrogen 14.0</div>	<div>8 <b>O</b> Oxygen 16.0</div>	<div>9 <b>F</b> Fluorine 19.0</div>	<div>10 <b>Ne</b> Neon 20.2</div>										
Period 3	<div>11 <b>Na</b> Sodium 23.0</div>	<div>12 <b>Mg</b> Magnesium 24.3</div>																	<div>13 <b>Al</b> Aluminum 27.0</div>	<div>14 <b>Si</b> Silicon 28.1</div>	<div>15 <b>P</b> Phosphorus 31.0</div>	<div>16 <b>S</b> Sulfur 32.1</div>	<div>17 <b>Cl</b> Chlorine 35.5</div>	<div>18 <b>Ar</b> Argon 39.9</div>										
Period 4	<div>19 <b>K</b> Potassium 39.1</div>	<div>20 <b>Ca</b> Calcium 40.1</div>	<div>21 <b>Sc</b> Scandium 45.0</div>	<div>22 <b>Ti</b> Titanium 47.9</div>	<div>23 <b>V</b> Vanadium 50.9</div>	<div>24 <b>Cr</b> Chromium 52.0</div>	<div>25 <b>Mn</b> Manganese 54.9</div>	<div>26 <b>Fe</b> Iron 55.8</div>	<div>27 <b>Co</b> Cobalt 58.9</div>	<div>28 <b>Ni</b> Nickel 58.7</div>	<div>29 <b>Cu</b> Copper 63.5</div>	<div>30 <b>Zn</b> Zinc 65.4</div>	<div>31 <b>Ga</b> Gallium 69.7</div>	<div>32 <b>Ge</b> Germanium 72.6</div>	<div>33 <b>As</b> Arsenic 74.9</div>	<div>34 <b>Se</b> Selenium 79.0</div>	<div>35 <b>Br</b> Bromine 79.9</div>	<div>36 <b>Kr</b> Krypton 83.8</div>																
Period 5	<div>37 <b>Rb</b> Rubidium 85.5</div>	<div>38 <b>Sr</b> Strontium 87.6</div>	<div>39 <b>Y</b> Yttrium 88.9</div>	<div>40 <b>Zr</b> Zirconium 91.2</div>	<div>41 <b>Nb</b> Niobium 92.9</div>	<div>42 <b>Mo</b> Molybdenum 95.9</div>	<div>43 <b>Tc</b> Technetium (98)</div>	<div>44 <b>Ru</b> Ruthenium 101.1</div>	<div>45 <b>Rh</b> Rhodium 102.9</div>	<div>46 <b>Pd</b> Palladium 106.4</div>	<div>47 <b>Ag</b> Silver 107.9</div>	<div>48 <b>Cd</b> Cadmium 112.4</div>	<div>49 <b>In</b> Indium 114.8</div>	<div>50 <b>Sn</b> Tin 118.7</div>	<div>51 <b>Sb</b> Antimony 121.8</div>	<div>52 <b>Te</b> Tellurium 127.6</div>	<div>53 <b>I</b> Iodine 126.9</div>	<div>54 <b>Xe</b> Xenon 131.3</div>																
Period 6	<div>55 <b>Cs</b> Cesium 132.9</div>	<div>56 <b>Ba</b> Barium 137.3</div>	<div>57 <b>La</b> Lanthanum 138.9</div>	<div>72 <b>Hf</b> Hafnium 178.5</div>	<div>73 <b>Ta</b> Tantalum 180.9</div>	<div>74 <b>W</b> Tungsten 183.8</div>	<div>75 <b>Re</b> Rhenium 186.2</div>	<div>76 <b>Os</b> Osmium 190.2</div>	<div>77 <b>Ir</b> Iridium 192.2</div>	<div>78 <b>Pt</b> Platinum 195.1</div>	<div>79 <b>Au</b> Gold 197.0</div>	<div>80 <b>Hg</b> Mercury 200.6</div>	<div>81 <b>Tl</b> Thallium 204.4</div>	<div>82 <b>Pb</b> Lead 207.2</div>	<div>83 <b>Bi</b> Bismuth 209.0</div>	<div>84 <b>Po</b> Polonium (209)</div>	<div>85 <b>At</b> Astatine (210)</div>	<div>86 <b>Rn</b> Radon (222)</div>																
Period 7	<div>87 <b>Fr</b> Francium (223)</div>	<div>88 <b>Ra</b> Radium (226)</div>	<div>89 <b>Ac</b> Actinium (227)</div>	<div>104 <b>Rf</b> Rutherfordium (261)</div>	<div>105 <b>Db</b> Dubnium (262)</div>	<div>106 <b>Sg</b> Seaborgium (266)</div>	<div>107 <b>Bh</b> Bohrium (264)</div>	<div>108 <b>Hs</b> Hassium (277)</div>	<div>109 <b>Mt</b> Meitnerium (268)</div>	<div>110 <b>Ds</b> Darmstadtium (271)</div>	<div>111 <b>Uus</b> Ununium (272)</div>	<div>112 <b>Uub</b> Unubium (285)</div>	<div>113 <b>Uut</b> Ununtrium (284)</div>	<div>114 <b>Uuq</b> Ununquadium (289)</div>	<div>115 <b>Uup</b> Ununpentium (288)</div>																			
<div>This zigzag line reminds you where the metals, nonmetals, and metalloids are.</div>																																		

This zigzag line reminds you where the metals, nonmetals, and metalloids are.

**Background**

Metals

Metalloids

Nonmetals

**Chemical symbol**

Solid

Liquid

Gas

A row of elements is called a **period**.

A column of elements is called a **group** or **family**.

Values in parentheses are the mass numbers of those radioactive elements' most stable or most common isotopes.

The discovery of elements 113, 114, and 115 has been reported but not confirmed.

The names and three-letter symbols of elements are temporary. They are based on the atomic numbers of the elements. Official names and symbols will be approved by an international committee of scientists.

<b>Lanthanides</b>	<b>58</b> <b>Ce</b> Cerium 140.1	<b>59</b> <b>Pr</b> Praseodymium 140.9	<b>60</b> <b>Nd</b> Neodymium 144.2	<b>61</b> <b>Pm</b> Promethium (145)	<b>62</b> <b>Sm</b> Samarium 150.4	<b>63</b> <b>Eu</b> Europium 151.9	<b>64</b> <b>Gd</b> Gadolinium 157.3	<b>65</b> <b>Tb</b> Terbium 158.9	<b>66</b> <b>Dy</b> Dysprosium 162.5	<b>67</b> <b>Ho</b> Holmium 164.9	<b>68</b> <b>Er</b> Erbium 167.3	<b>69</b> <b>Tm</b> Thulium 168.9	<b>70</b> <b>Yb</b> Ytterbium 173.0	<b>71</b> <b>Lu</b> Lutetium 175.0
<b>Actinides</b>	<b>90</b> <b>Th</b> Thorium 232.0	<b>91</b> <b>Pa</b> Protactinium 231.0	<b>92</b> <b>U</b> Uranium 238.0	<b>93</b> <b>Np</b> Neptunium (237)	<b>94</b> <b>Pu</b> Plutonium (244)	<b>95</b> <b>Am</b> Americium (243)	<b>96</b> <b>Cm</b> Curium (247)	<b>97</b> <b>Bk</b> Berkelium (247)	<b>98</b> <b>Cf</b> Californium (251)	<b>99</b> <b>Es</b> Einsteinium (252)	<b>100</b> <b>Fm</b> Fermium (257)	<b>101</b> <b>Md</b> Mendelevium (258)	<b>102</b> <b>No</b> Nobelium (259)	<b>103</b> <b>Lr</b> Lawrencium (262)

These elements are placed below the table to allow the table to be narrower.

# Section Exploring The Periodic Table Holt Science Spectrum

**Adrian Dingle, Dan Green**



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Exploring the Elements Isabel Thomas,2020 Science meets design in this comprehensive introduction to the chemical elements that make up our universe This artful and accessible guide to the periodic table the ultimate reference tool for scientists worldwide names all 118 chemical elements and helps young readers understand the remarkable ways we have learned to use them Graphically stunning layouts feature each element s letter symbol and atomic number exploring its attributes characteristics uses and interesting stories behind its discovery Complete with a comprehensive introduction conclusion and glossary this is the perfect introduction to chemistry for inquisitive minds

Elements Theodore Gray,2012-04-03 With more than 1 million copies sold worldwide The Elements is the most entertaining comprehensive and visually arresting book on all 118 elements in the periodic table Includes a poster of Theodore Gray s iconic photographic periodic table of the elements Based on seven years of research and photography by Theodore Gray and Nick Mann The Elements presents the most complete and visually arresting representation available to the naked eye of every atom in the universe Organized sequentially by atomic number every element is represented by a big beautiful photograph that most closely represents it in its purest form Several additional photographs show each element in slightly altered forms or as used in various practical ways Also included are fascinating stories of the elements as well as data on the properties of each including atomic number atomic symbol atomic weight density atomic radius as well as scales for electron filling order state of matter and an atomic emission spectrum This of solid science and stunning artistic photographs is the perfect gift book for every sentient creature in the universe

Holt Science and Technology Holt, Rinehart and Winston Staff,2003-01

The Periodic Table Adrian Dingle,Dan Green,2014 The original Basher Science made even better

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