



Science Curriculum

Chemical Changes

Read the passage and answer the following questions.



1. What is the chemical formula for hydrogen peroxide?

Hydrogen peroxide is the chemical formula for the breaking of molecules. In this chemical reaction, the reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2). The reaction is exothermic, meaning it releases heat and energy.

2. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

3. What is the chemical formula for oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product?

Hydrogen peroxide is a colorless liquid with a strong, pungent odor. It is a powerful oxidizing agent and is used in many industrial and household applications. In this reaction, it is broken down into water and oxygen. The reaction is exothermic, meaning it releases heat and energy. The reaction is also catalyzed by certain substances, such as manganese dioxide, which speeds up the reaction without being consumed.

4. According to the law of conservation of mass, if 100g of hydrogen peroxide is decomposed, how much water and oxygen will be produced? Is this a chemical or a physical change? Is it reversible?

100 grams



5. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product?

6. What is the chemical formula for oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product?

7. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product?

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9. What is the chemical formula for oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product?

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10. According to the law of conservation of mass, if 100g of hydrogen peroxide is decomposed, how much water and oxygen will be produced? Is this a chemical or a physical change? Is it reversible?

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McGraw-Hill Staff



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