

Net Ionic Equations Worksheet

Write balanced chemical equations for dissolving the following compounds in water:

1. Solid sodium chloride



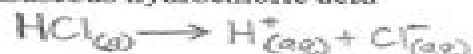
2. Solid barium nitrate



3. Solid magnesium hydroxide



4. Gaseous hydrochloric acid



5. Solid calcium phosphate



6. Solid iron (III) sulfide



7. Gaseous carbon dioxide (*Hint: Carbon dioxide will form carbonic acid first then decomposes into ions*)



Write balanced chemical equations and then net ionic equations for the following reactions:

8. Aqueous sodium sulfate and aqueous barium chloride



9. Aqueous lithium bromide and aqueous silver nitrate



10. Aqueous lead II acetate and aqueous sodium carbonate



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Holt Rinehart & Winston



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