

Endothermic and Exothermic reaction Worksheet

Name _____ date _____ period _____

1 Exothermic and endothermic reactions

Decide whether each of these reactions is exothermic or endothermic:

- a) When two chemicals mix their temperature rises: ____exo____
- b) A solid burns brightly and releases heat, light and sound: ____exo____
- c) When two chemicals are mixed their temperature drops: ____endo____
- d) Two chemicals will only react if you heat them continually: ____endo____
- e) Plants take in light energy for photosynthesis: ____endo____

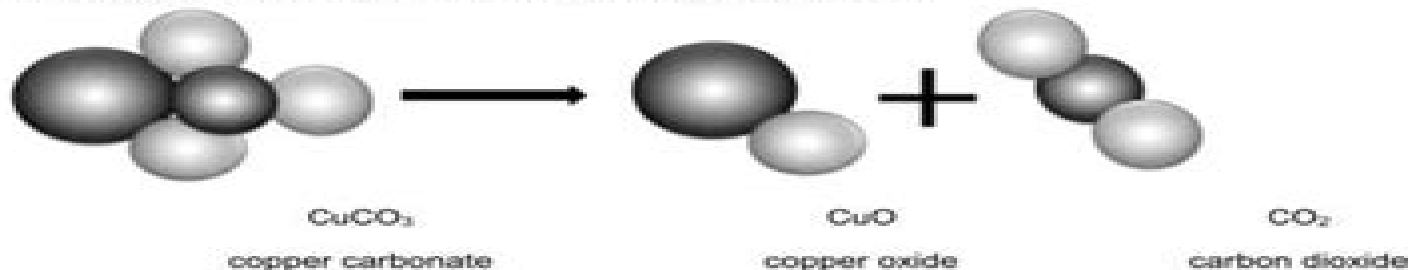
2 Making and breaking bonds

During chemical reactions the bonds between atoms break and new bonds form.

Energy must be absorbed to break a bond, so breaking bonds is endothermic.

Making new bonds is exothermic because energy is released.

- a) When green copper carbonate decomposes, the equation is:



Is the reaction exothermic or endothermic? Use ideas about bonds to explain why.

__exo because two bonds are broken from 4 bonds

- b) Draw diagrams to show what happens when hydrogen reacts with oxygen. Mark the bonds broken in blue and the new bonds formed in red. The equation is:



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**Thirumalaisamy P.
Velavancorresponding**



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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Radioactivity And Nuclear Reactions Page 35 Answer Key**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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