

Radioactive Dating Worksheet

Below are several problems involving radioactive dating. Use the half-life of the radioactive element to determine the age of the sample. Show your work for each problem.

1. A sample of carbon-14 is found to have a mass of 10.0 grams. After 10,000 years, the mass is found to be 6.25 grams. How old is the sample? (Half-life of carbon-14 is 5,730 years.)

$$N = N_0 \left(\frac{1}{2} \right)^{\frac{t}{t_{1/2}}}$$

$$6.25 = 10.0 \left(\frac{1}{2} \right)^{\frac{t}{5730}}$$

2. A sample of uranium-238 is found to have a mass of 10.0 grams. After 10,000 years, the mass is found to be 6.25 grams. How old is the sample? (Half-life of uranium-238 is 4,468 years.)

$$N = N_0 \left(\frac{1}{2} \right)^{\frac{t}{t_{1/2}}}$$

$$6.25 = 10.0 \left(\frac{1}{2} \right)^{\frac{t}{4468}}$$

Because the half-life of uranium-238 is 4,468 years, after 10,000 years, only 6.25% of the original sample remains. This is the same as saying that 93.75% of the sample has decayed.



3. A sample of carbon-14 is found to have a mass of 10.0 grams. After 10,000 years, the mass is found to be 6.25 grams. How old is the sample? (Half-life of carbon-14 is 5,730 years.)

Use the same method as in problem 1 to determine the age of the sample.

Radiometric Dating Worksheet Answer Key

Thomas Lundqvist

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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