

Human Impacts on the Earth Environment

Acid Rain

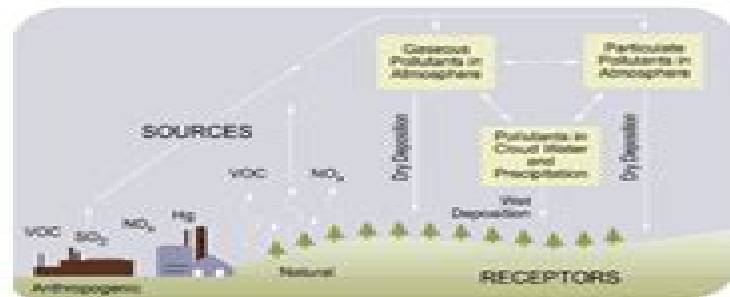
Acid Rain

- Buildings such as power plants or refineries expel sulfur and nitrogen oxides into the air, which combine with water vapor to produce sulfuric and nitric acids.
- These acids in the atmosphere then are deposited onto the environment in the form of acid rain. It can also be deposited as fog, snow, or particulates (in dry areas).
- Acid rain** typically has a pH of less than 5.0, whereas regular rain has a pH of around 5.6.
- The **pH scale** measures the acidity of a substance, the lower the pH value, the more acidic a substance is (more **alkaline**).
- Acid rain is typically found in industrialized areas

Study Tip

To remember the pollutants which cause acid rain:

SNAP: Sulfur Nitrogen Are Pollutants



Sulfuric acid, nitric acid, and mercury are among the many types of pollutants that cause acid rain

The Effects of Acid Rain

- Acid rain destroys plants and makes them more vulnerable to destruction.
- Some acids deposit on the ground dry, resulting in the soil being stripped of its basic nutrients.
- Small animals often die when they are exposed to acid rain, causing larger animals to lose a valuable food source.
- Fish die in water that is too acidic, and crops cannot grow in soil exposed to acid rain.
- Ruins buildings
- Since acid rain can travel far, its effects can be spread to far distances.



Acid rain can deteriorate trees and other plants as well.

Concept Check

- Discuss how acid forms, beginning with the formation of acid from refineries or power plants
- What is acid rain and why is it bad?

Raining Study Guide Answer

D Kirk



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 - Advantages of eBooks Over Traditional Books
2. Identifying Raining Study Guide Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Raining Study Guide Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Raining Study Guide Answer
 - Personalized Recommendations
 - Raining Study Guide Answer User Reviews and Ratings
 - Raining Study Guide Answer and Bestseller Lists
5. Accessing Raining Study Guide Answer Free and Paid eBooks
 - Raining Study Guide Answer Public Domain eBooks

- Raining Study Guide Answer eBook Subscription Services
- Raining Study Guide Answer Budget-Friendly Options
- 6. Navigating Raining Study Guide Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Raining Study Guide Answer Compatibility with Devices
 - Raining Study Guide Answer Enhanced eBook Features
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
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- 9. Balancing eBooks and Physical Books Raining Study Guide Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Raining Study Guide Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Raining Study Guide Answer
 - Setting Reading Goals Raining Study Guide Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Raining Study Guide Answer
 - Fact-Checking eBook Content of Raining Study Guide Answer
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

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- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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