

EXERCISE - AQUEOUS SOLUTIONS AND WATER QUALITY

As you read through pages 357-362, answer the following questions:

- 1) What % of the world's fresh water does Canada have?
- 2) Look at the data in Table 9.5. Did you realize that it is acceptable to have small amounts of substances such as uranium, mercury and insecticides in drinking water?
- 3) a) What type of bacteria in the water supply caused 7 people to die and many to become ill in Walkerton, Ont. in 2000?
b) What should the town's water manager have done to deal with this problem?

Sources That Compromise Water Quality

- 1) State three examples of point sources of pollution.
- 2) State one example of a diffuse source of pollution.
- 3) State one example of an indirect source of pollution.

Treating Water for Your Home

- 1) Briefly describe the six processes that take place at a water treatment plant.
- 2) Why do some municipalities add sodium fluoride to drinking water?

Hard Water and Soft Water/Treating Water at Home

- 1) Distinguish between hard water and soft water.
- 2) What is the most common type of rock that causes hard water?
- 3) Where is boiler scale commonly found and what problems does it cause?
- 4) a) In an ion exchange water softener, what ion exchanges with the calcium and other hardness causing ions in the water?
b) Why is this kind of water softener not the best choice for some people?

Waste-Water Treatment

What does each of the following mainly involve:

- 1) primary waste-water treatment?
- 2) secondary waste-water treatment?
- 3) tertiary waste-water treatment?

Solution Stoichiometry

Read The Concentration of Ions Sample Problem on pages 348-349.
Then, do question 8 on page 352.
(You can check your answer to this question on page 367.)

Water And Aqueous Systems Vocabulary Review Answer Key

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Table of Contents Water And Aqueous Systems Vocabulary Review Answer Key

1. Understanding the eBook Water And Aqueous Systems Vocabulary Review Answer Key
 - The Rise of Digital Reading Water And Aqueous Systems Vocabulary Review Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Water And Aqueous Systems Vocabulary Review Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Water And Aqueous Systems Vocabulary Review Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Water And Aqueous Systems Vocabulary Review Answer Key
 - Personalized Recommendations
 - Water And Aqueous Systems Vocabulary Review Answer Key User Reviews and Ratings
 - Water And Aqueous Systems Vocabulary Review Answer Key and Bestseller Lists
5. Accessing Water And Aqueous Systems Vocabulary Review Answer Key Free and Paid eBooks
 - Water And Aqueous Systems Vocabulary Review Answer Key Public Domain eBooks
 - Water And Aqueous Systems Vocabulary Review Answer Key eBook Subscription Services
 - Water And Aqueous Systems Vocabulary Review Answer Key Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - Water And Aqueous Systems Vocabulary Review Answer Key Compatibility with Devices
 - Water And Aqueous Systems Vocabulary Review Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Water And Aqueous Systems Vocabulary Review Answer Key
 - Highlighting and Note-Taking Water And Aqueous Systems Vocabulary Review Answer Key
 - Interactive Elements Water And Aqueous Systems Vocabulary Review Answer Key
8. Staying Engaged with Water And Aqueous Systems Vocabulary Review Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Water And Aqueous Systems Vocabulary Review Answer Key
9. Balancing eBooks and Physical Books Water And Aqueous Systems Vocabulary Review Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Water And Aqueous Systems Vocabulary Review Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Water And Aqueous Systems Vocabulary Review Answer Key
 - Setting Reading Goals Water And Aqueous Systems Vocabulary Review Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Water And Aqueous Systems Vocabulary Review Answer Key
 - Fact-Checking eBook Content of Water And Aqueous Systems Vocabulary Review Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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