

Temperature, Thermal Energy, and Heat

Understanding Main Ideas

Answer the following questions in the space provided.

1. Explain how it is possible for objects at the same thermal energy, but having three different temperatures.

Ans. An object's ability to increase or fall in the amount of the amount of thermal energy it has depends on its mass.

2. Describe what is meant by absolute zero.

Absolute zero is the lowest possible temperature where the thermal energy of an object is zero.

3. Explain how different objects can have the same thermal energy, but have different temperatures.

Thermal energy is the total energy of an object, which is the sum of the kinetic and potential energies of all the particles in the object.

4. Compare the following units to make up the Celsius and the Fahrenheit temperature scales.

The Celsius scale is based on the freezing and boiling points of water, while the Fahrenheit scale is based on the freezing and boiling points of a mixture of water and alcohol.

5. Compare the changes in temperature of 1°C and 1°F.

A change of 1°C is equivalent to a change of 1.8°F.

Building Vocabulary

Match each term with its definition, using the lettered definitions in the right column. One letter will be used more than once.

16. a heating

17. b temperature

18. c heat

19. d thermal energy

- a. the total kinetic energy of all the particles in an object

- b. the amount of the average kinetic energy of the particles in an object

- c. the transfer of thermal energy from one object to another

- d. the total kinetic energy of all the particles in an object

Thermal Energy And Heat Review And Reinforce Pearson

Michal Rosen-Zvi



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