

Chapter 2.1 Chemistry in Living Systems – Chemical Bonds and Reactions Reading Guide

1. How do bombardier beetles use chemistry to maintain homeostasis and survive in their environment? Read paragraph on page 713
2. What in your digestive track helps you break down the food you eat?
3. More specifically, what things can HCl break down?
4. List the steps in the digestion process of interactions.
5. What do the salivary glands break down?
6. What 2 things in the stomach help break down food quickly?
7. Describe what the term macroscopic means?
8. All organisms depend on different _____ and _____ reactions.
9. Define an atom.
10. How many atoms fit in a space the size of a period?
11. True or False: All atoms have different basic structures.
12. Define an element.
13. Refer to Figure 2 to fill in the missing components of the subatomic particles in an atom.

Reading Guide Reactions

Jiben Roy



Reading Guide Reactions:

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web aug 19 2022 atoms with the same number of protons but different numbers of neutrons are called isotopes they share almost the same chemical properties but differ in mass and therefore in physical properties there are stable isotopes which do not emit radiation and there are unstable isotopes which do emit radiation

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web oct 31 2023 an isotope is one of two or more species of atoms of a chemical element with the same atomic number and position in the periodic table and nearly identical chemical behavior but with different atomic masses and physical properties

every chemical element has one or more isotopes

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