

# Algebra Practice

## Adding & Subtracting Polynomials

*You can do your rough work on a separate paper, or use algebra tiles to help you if you like!*

1) Add each set of polynomials. Simplify your answer. Choose an answer from the answer bank below and drag it to the correct box.

a)  $(2x + 1) + (3x + 8) =$  \_\_\_\_\_

b)  $(4x + 9) + (7x - 3) =$  \_\_\_\_\_

c)  $(2x - 6) + (-5x + 1) =$  \_\_\_\_\_

d)  $(4x^2 - 6x + 9) + (x^2 - 2) =$  \_\_\_\_\_

e)  $(x^2 - 2x + 1) + (3x^2 - 5x - 2) =$  \_\_\_\_\_

Answer bank \*\* Note there are more answers than questions \*

$5x^2 + 6x + 7$	$11x - 6$	$5x^2 - 6x + 7$
$5x^2 + 6x - 7$	$11x + 6$	$-3x - 5$
$4x^2 - 7x - 1$	$5x + 9$	$-3x - 5x$
$4x^2 - 8x$	$5x + 7$	$4x^2 + 3x - 1$

# Skills Practice 8 5 Adding And Subtracting Polynomials

**Caren Van Slyke**



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web this is like a one dimensional system whose mechanical energy  $e$  is a constant and whose potential energy with respect to zero energy at zero displacement from the spring  $s$  unstretched length  $x = 0$  is  $u = \frac{1}{2} kx^2$  figure 8 5 2 8 5 2 a a glider between springs on an air track is an example of a horizontal mass spring system

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web 1 if the pe of the reactants is 430kj and the pe of the products is 190kj what is the  $\Delta h$  2 if the pe of the reactants is 430kj and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed image text potential energy diagram answer the questions using the graph

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in an endothermic reaction the energy of the products is greater than the energy of the reactants and  $\Delta h$  is positive b in an exothermic reaction the energy of the products is lower than the energy of the

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate  $\Delta h$  for the reaction and label it with the correct value see attached image

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web use the potential energy diagram to answer the questions below 1 is the overall reaction as shown exothermic or endothermic 2 what is the activation energy for the forward reaction 3 what is the