

# Graphing Periodic Trends Activity


Name: \_\_\_\_\_ Period: \_\_\_\_\_

Use the information in these tables to complete the graph as described below

Symbol	Atomic Radius (Picometers)	First Ionization energy (kJoules/mole)	Electronegativity (4-point scale)
H	31	1312	2.1
Li	128	520	1.0
Na	166	496	0.9
K	203	410	0.8
Rb	220	403	0.8
Cs	244	376	0.7

Symbol	Atomic Radius (Picometers)	First Ionization energy (kJoules/mole)	Electronegativity (4-point scale)
Na	166	496	0.9
Mg	141	738	1.2
Al	121	578	1.5
Si	111	787	1.8
P	107	1012	2.1
S	105	1000	2.5
Cl	102	1251	3.0
Ar	106	1521	---

1. Using colored pencils or pen, list under the symbol (in this order!) the Atomic Radius, First Ionization Energy, and Electronegativity.
2. Use a different color for each property. Example: Write all of the Atomic radius values in red, all of the First ionization energies in green, and all of the Electronegativities in blue. Of course, you can pick any colors that are available, as long as you are consistent.
3. Observe the trends in each property as you go down the Alkali metal group, and as you go across Period 3.
4. Write out each of the statements written on the opposite side, completing each statement with the observed trend (increase or decrease)

H	<p>Across a period (→) atomic radius tends to _____.</p> <p>Across a period (→) first ionization energy tends to _____.</p> <p>Across a period (→) electronegativity tends to _____.</p> 						
Li							
Na 166 496 0.9							
K	<p>Down a Group (↓) atomic radius tends to _____.</p> <p>Down a Group (↓) first ionization energy tends to _____.</p> <p>Down a Group (↓) electronegativity tends to _____.</p>						
Rb							
Cs							

# Trends In The Periodic Table Worksheet Holt

**Yan Bai**



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Holt Chemistry Holt Rinehart & Winston, 2003-01-24

*Mastering the Periodic Table* Cybellium Ltd, 2024-10-26

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*Periodic Table, The: Past, Present, And Future* Geoff Rayner-canham, 2020-08-04 This is an an absolutely wonderful book that is full of gems about the elements and the periodic table All in all the book is highly recommended to philosophers of chemistry As philosophers we have a natural tendency to concentrate on generalities and not to get too involved in the specifics and the details Above all else this new book reminds us that such an approach needs to be tempered by a detailed knowledge of the exceptions and features that go against the simplified generalities which we so cherish Read Full Review Eric Scerri Foundations of Chemistry Many questions are dealt with in a clearly written way in this stimulating and innovative book The reader will quickly become interested in the subject and will be taken on tour through this Periodic Table in a very readable way both for students and teachers The number of illustrations is good and clear This book is indeed unique and quite thought provoking This book is highly recommended for students teachers researchers and not only chemists Geologists biochemist and also physicists will find it very interesting to read Read Full Review Chemistry International That fossilized chart on every classroom wall isn t that The Periodic Table Isn t that what Mendel ev devised about a century ago No and No There are many ways of organizing the chemical elements some of which are thought provoking and which reveal philosophical challenges Where does hydrogen belong Can an element occupy more than one location on the chart Which are the Group 3 elements Is aluminum in the wrong place Why is silver I like thallium I Why is vanadium like molybdenum Why does gold form an auride ion like a halide ion Does an atom know if it is a non metal or metal Which elements are the metalloids Which are the triels So many questions In this stimulating and innovative book the Reader will be taken on a voyage from the past to the present to the future of the Periodic Table This book is unique This book is readable This book is thought provoking It is a multi dimensional examination of patterns and trends among the chemical elements Every reader will discover something about the chemical elements which will provoke thought and a new appreciation as to how the elements relate together Understanding the Periodic Table: A Chemistry Guide Cybellium, Welcome to the forefront of knowledge with Cybellium your trusted partner in mastering the cutting edge fields of IT Artificial Intelligence Cyber

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*Chemistry and Our Universe* , Return to the periodic table introduced in Lecture 1 to practice predicting properties of elements based on their electronic structure Then witness what happens when three different alkali metals react with water Theory forecasts a pronounced difference in the result Is there

**The Periodic Table** Eric R. Scerri, 2006-10-12 The periodic table is one of the most potent icons in science It lies at the core of chemistry and embodies the most fundamental principles of the field The one definitive text on the development of the periodic table by van Spronsen 1969 has been out of print for a considerable time The present book provides a successor to van Spronsen but goes further in giving an evaluation of the extent to which modern physics has or has not explained the periodic system The book is written in a lively style to appeal to experts and interested lay persons alike The Periodic Table begins with an overview of the importance of the periodic table and of the elements and it examines the manner in which the term element has been interpreted by chemists and philosophers The book then turns to a systematic account of the early developments that led to the classification of the elements including the work of Lavoisier Boyle and Dalton and Cannizzaro The precursors to the periodic system like D bereiner and Gmelin are discussed In chapter 3 the discovery of the periodic system by six independent scientists is examined in detail Two chapters are devoted to the discoveries of Mendeleev the leading discoverer including his predictions of new elements and his accommodation of already existing elements Chapters 6 and 7 consider the impact of physics including the discoveries of radioactivity and isotopy and successive theories of the electron including Bohr s quantum theoretical approach Chapter 8 discusses the response to the new physical theories by chemists such as Lewis and Bury who were able to draw on detailed chemical knowledge to correct some of the early electronic configurations published by Bohr and others Chapter 9 provides a critical analysis of the extent to which modern quantum mechanics is or is not able to explain the periodic system from first principles Finally chapter 10 considers the way that the elements evolved following the Big Bang and in the interior of stars The book closes with an examination of further chemical aspects including lesser known trends within the periodic system such as the knight s move relationship and secondary periodicity as well as attempts to explain such trends

*Trends in the Periodic Table* Open University. S25-Course Team, 1972 Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 Holt Science

and Technology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-01-01

## **Holt Science Spectrum**

**Physical Science Chapter 5 Resource File: the Periodic Table** Holt Rinehart & Winston, 2008-01-01 *The Periodic*

*Table* Eric R. Scerri, 2019-07-25 The periodic table of elements first encountered by many of us at school provides an arrangement of the chemical elements ordered by their atomic number electron configuration and recurring chemical properties and divided into periodic trends In this Very Short Introduction Eric R Scerri looks at the trends in properties of elements that led to the construction of the table and shows how the deeper meaning of the table s structure gradually became apparent with the development of atomic theory and in particular quantum mechanics which underlies the behaviour of all of the elements and their compounds This new edition publishing in the International Year of the Periodic Table celebrates the completion of the seventh period of the table with the ratification and naming of elements 113 115 117 and 118 as nihonium moscovium tennessine and oganesson Eric R Scerri also incorporates new material on recent advances in our understanding of the origin of the elements as well as developments concerning group three of the periodic table ABOUT THE SERIES The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area These pocket sized books are the perfect way to get ahead in a new subject quickly Our expert authors combine facts analysis perspective new ideas and enthusiasm to make interesting and challenging topics highly readable

**Chemical Periodicity** Robert Thomas Sanderson, 2013-04-20

## **New Ideas in Chemistry from Fresh Energy for the**

**Periodic Law** Henry Bent, 2006-09-05 New Ideas calls to mind Aristotle s synopsis of the Iliad and the Odyssey Woman abducted Long war One guy has a hard time getting home End of story The rest is episodes Similarly here Chemical capture of the Left Step Periodic Table One element finds a new home The noblest of the noble gases is not a Noble Gas End of story The rest is novel consequences of the Noble Gas Conclusion Among them overlooked Rules of Triads Block Sizes and Full Shells overlooked block to block trends and a correspondence between elements ordinal numbers in their Groups and orbital s radial quantum numbers and recognition that Pauli s explanation of Periodicity s magic numbers 2 8 18 got the right answer the Pauli Exclusion Principle for the wrong reason New Ideas ends with suggestions for streamlining the teaching of the mole concept chemical bonding and thermodynamics in order to provide room in the chemistry curriculum for a more thorough treatment of Periodic System Systematics

## **Science Comics: The Periodic Table of Elements**

Jon Chad, 2023-03-14 Step out of your element with Science Comics The Periodic Table of Elements the latest volume of First Second s nonfiction graphic novel series A book of fun chemistry experiments has fallen into the wrong hands Only Mel can use her knowledge of the periodic table to put an end to a maniacal madman s evil schemes The periodic table helps us quickly understand the 118 elements those tiny substances that make up everything in the world By using the periodic table we can recognize how these building blocks behave find trends and patterns in the universe and make predictions about elements that haven t been discovered yet Join us in learning about the periodic table and maybe the next big discovery will

be yours      **Holt Science and Technology** Holt, Rinehart and Winston Staff,2003-01      **Periodic Table: A Formula Handbook** N.B. Singh, Periodic Table A Formula Handbook is a concise and indispensable guide to the elements providing a comprehensive collection of essential formulas properties and trends within the periodic table This handbook equips students scientists and enthusiasts with quick access to vital information on each element including atomic number atomic mass electron configuration and chemical properties With clear organization and easy to understand explanations this book serves as an invaluable resource for anyone studying chemistry conducting research or simply seeking to deepen their understanding of the fundamental building blocks of matter      *The Periodic Table Book* DK,2017-03-30 This eye popping encyclopedia takes you on a tour of all the world s elements From argon to zinc each and every one of the 118 chemical elements are explored in dazzling detail With the periodic table celebrating its 150th anniversary in 2019 you ll be in your element as you discover the incredible variety of building blocks that make up our Universe and learn the remarkable ways we now use them More than 1 000 photographs showcase the natural forms of each element and the range of everyday and unusual objects where they can be seen This helps children understand exactly where the different elements have found their place in the world The true science behind the elements is explained in properties atomic structure and table position This essential book turns the tables on traditional reference and presents the periodic table as never before to appeal to school children today Included is a giant glossy pull out poster perfect to aid classroom discussions or to dazzle on bedroom walls

### **TRENDS IN THE PERIODIC TABLE OF THE ELEMENTS COMPUTER BASED INSTRUCTION IN CHEMISTRY.**

,1992      [Atoms, Molecules & Elements: The Periodic Table Gr. 5-8](#) George Graybill,2015-10-01 This is the chapter slice The Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives      **Periodic Table (Advanced)** Speedy Publishing,2014-08-01 An advanced periodic table of elements displays not only the elements but the ions that form each element A pamphlet with such a visual aid would greatly benefit chemistry students Any student taking chemistry will need to learn the elements A pamphlet would be concise and break the information down simply making it easier to understand and remember It allows students to simply focus on the main point rather than taking in information that they may or may not need

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