

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____

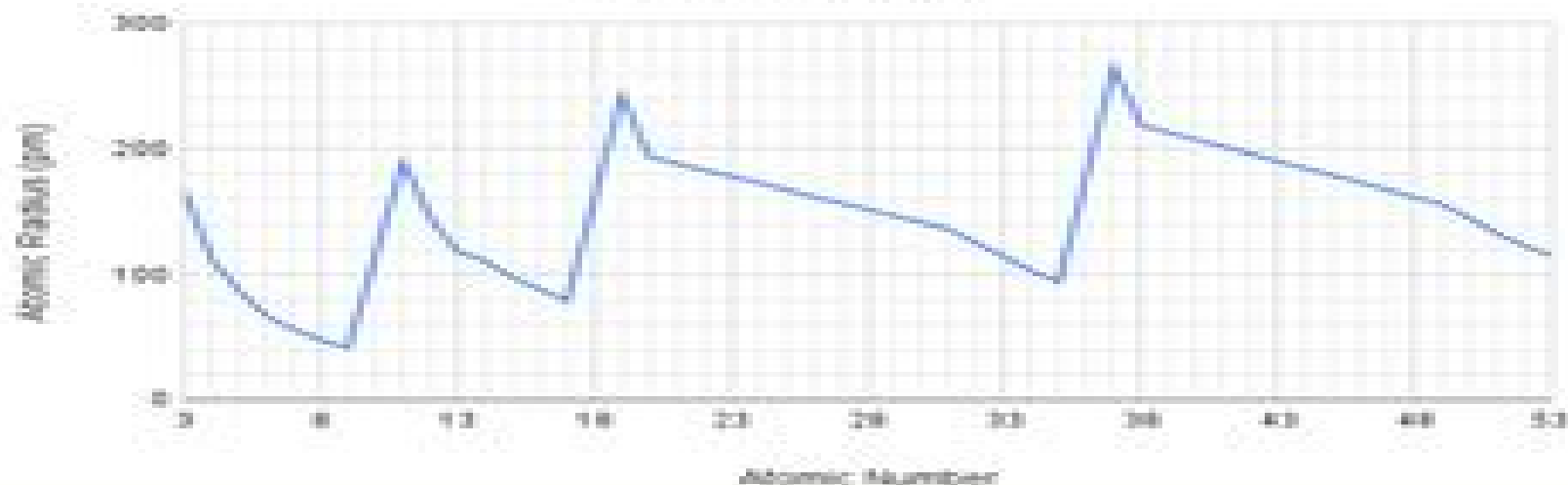
Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	152	Sodium - 11	190	Potassium - 19	243	Rubidium - 37	265
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 38	219
Boron - 5	87	Aluminum - 13	118	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 32	125	Tin - 50	145
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	115

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 37	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 38	550

Atomic Radius



Trends In The Periodic Table Graphing Answers

Randy L. Bell, Joe Garofalo



Trends In The Periodic Table Graphing Answers:

Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. WileyPLUS sold separately from text. *The Science Teachers Bulletin*, 1989. Science Units for Grades 9-12 Randy L. Bell, Joe Garofalo, 2005. Sample topics include cell division, virtual dissection, earthquake modeling, the Doppler Effect and more.

Jacaranda Chemistry 1 VCE Units 1 and 2, learnON and Print Neale Taylor, Angela Stubbs, Robert Stokes, 2022-11-30. Developed by expert Victorian teachers for VCE students. The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities. Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023/2027. CBSE Most Likely Question Bank Science Class 10 (2022 Exam) - Categorywise & Chapterwise with New Objective Paper Pattern, Reduced Syllabus Gurukul, 2021-06-15. Benefit from Chapter Wise Section wise Question Bank Series for Class 10 CBSE Board Examinations 2022 with our Most Likely CBSE Question Bank for Science having Physics Chemistry and Biology Subject Wise books designed to prepare and practice effectively each subject at a time. Our Most Probable Question Bank highlights the knowledge based and skill based questions such as Summary MCQs, Reasoning Based Questions, Very Short Questions, Formula Based Questions, Short Questions, Diagram Based Questions, Differentiate Between Analysis and Evaluation Based Questions, Practical Based Questions, Numericals, Assertion and Reasoning Based Questions, Creating Based Questions, Case Based Questions and Test Your Knowledge. Our handbook will help you study and practice well at home. How can you benefit from Gurukul Most Likely CBSE Science Question Bank for 10th Class? Our handbook is strictly based on the latest syllabus prescribed by the council and is categorized chapterwise/topicwise to provide in depth knowledge of different concept questions and their weightage to prepare you for Class 10th CBSE Board Examinations 2022. 1. Focussed on New Objective Paper Pattern Questions. 2. Includes Solved Board Exam Paper 2020 for both Delhi and outside Delhi Set 1, 3 and Toppers Answers 2019. 3. Previous Years Board Question Papers Incorporated. 4. Visual Interpretation as per latest CBSE Syllabus. 5. Exam Oriented Effective Study Material provided for Self Study. 6. Chapter Summary for Easy Quick Revision. 7. Having frequently asked questions from Compartment Paper, Foreign Paper and latest Board Paper. 8. Follows the Standard Marking

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Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life

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Principles of Epidemiology Workbook: Exercises and Activities Ray M. Merrill, 2010-09-15 This workbook was written for students of epidemiology and serves as a supplement to any one of several introductory text books in epidemiology Each chapter is divided into an introduction a series of questions and detailed responses and a series of Homework questions At the end of each chapter is a table with a list of selected epidemiology text books with accompanying chapters in those books that the workbook chapter may supplement The general learning outcomes LOs for this workbook are

- 1 Become familiar with basic concepts and definitions commonly used in epidemiology
- 2 Define a public health problem
- 3 Identify appropriate uses and limitations of data and research design strategies for solving public health problems
- 4 Make relevant inferences from quantitative and qualitative data
- 5 Distinguish between statistical association and cause effect relationships
- 6 Measure and describe patterns of disease incidence prevalence and mortality
- 7 Identify environmental factors and behaviors associated with health related states or events
- 8 Be familiar with the steps for investigating disease outbreaks
- 9 Identify calculate and interpret common indices used in identifying the health status
- 10 Evaluate program effectiveness
- 11 Critically assess epidemiological research
- 12 Be able to communicate health findings

Each chapter features 10 20 mastery check questions

with detailed answers 5 optional problems A case study A multiple choice short answer quiz Answers to the cases and quizzes are provided as part of the online instructor resource package Jacaranda Core Science Stage 4 New South Wales

Australian Curriculum, 3e learnON and Print Paul Arena,2025-08-25 **Sif Chemistry NI Twb 2e** Rex M. Heyworth,2007

A-Level Practice Questions Chemistry Ed H2.2 CS Toh,2016-07-26 This is an ebook version of the A Level Practice Questions Chemistry Higher 2 Ed H2 2 published by Step by Step International Pte Ltd For the revised Higher 2 H2 syllabus with first exam in 2017 This ebook contains typical questions for readers to practise with It provides concise suggested solutions to illustrate the essential steps taken to apply the relevant theories and how the suggested answers are obtained We believe the suggested solutions will help readers learn to learn and apply the relevant knowledge The questions and suggested solutions are organised by topics to facilitate referring to them as the topics are being discussed

Understanding Advanced Physical Inorganic Chemistry: The Learner's Approach (Revised Edition) Kim Seng Chan,Jeanne Tan,2016-09-26 This revised edition has been updated to meet the minimum requirements of the new Singapore GCE A level syllabus that would be implemented in the year 2016 Nevertheless this book is also highly relevant to students who are studying chemistry for other examination boards In addition the authors have also included more Q A to help students better understand and appreciate the chemical concepts that they are mastering **Holt Chemistry** Ralph Thomas Myers,2004

Chemistry Therald Moeller,2012-12-02 Chemistry with Inorganic Qualitative Analysis is a textbook that describes the application of the principles of equilibrium represented in qualitative analysis and the properties of ions arising from the reactions of the analysis This book reviews the chemistry of inorganic substances as the science of matter the units of measure used atoms atomic structure thermochemistry nuclear chemistry molecules and ions in action This text also describes the chemical bonds the representative elements the changes of state water and the hydrosphere which also covers water pollution and water purification Water purification occurs in nature through the usual water cycle and by the action of microorganisms The air flushes dissolved gases and volatile pollutants when water seeps through the soil it filters solids as they settle in the bottom of placid lakes Microorganisms break down large organic molecules containing mostly carbon hydrogen nitrogen oxygen sulfur or phosphorus into harmless molecules and ions This text notes that natural purification occurs if the level of contaminants is not so excessive This textbook is suitable for both chemistry teachers and students

Exploring Physical Science in the Laboratory John T. Salinas ,2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging Exploring Physical Science in the Laboratory guides students through the mysteries of the observable world and helps them develop a

clear understanding of challenging concepts *Cracking the ACT, 2005 Edition* Princeton Review (Firm),2005-01-04 The country's leading test prep company offers students proven techniques and strategies for scoring high on the American College Testing Assessment Includes two sample tests *Educart CBSE Class 12 Final Revision Book 2025 - Physics + Chemistry+ Mathematics + English Core (2024-25)* Educart,2025-02-18 What You Get Ch wise Important Q's Sample Papers Educart CBSE Class 12 Final Revision Book 2025 Strictly based on sample papers released by CBSE for 2025 exam preparation Includes ch wise important questions for each of the four subjects Includes unit wise quick revisions for each of four subjects Practice questions from sample papers putting what you learnt to the test Why choose this book Best resource for structured and quick revision for the final board exams **New Coordinated Science: Chemistry Students' Book** RoseMarie Gallagher,R. Gallagher,Paul Ingram,2001-07-19 Provides information in manageable chunks which is reinforced by questions and activities that encourage students to consider the practical application of science to everyday life This work is useful for Higher Tier GCSE students

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