

**General, Organic & Biological Chemistry, 5e (Timberlake)**  
**Chapter 7 Chemical Reactions and Quantities**

**7.1 Multiple-Choice Questions**

1) A chemical equation is balanced when

- A) the total number of molecules is the same in reactants and products.
- B) the total number of ions is the same in reactants and products.
- C) the sum of the coefficients of the reactants is equal to the sum of the coefficients of the products.
- D) the number of atoms of each element is the same in reactants and products.
- E) the charge on each atom is the same in reactants and products.

Answer: D

2) In any balanced chemical equation, the number of each type of atom on both sides of the equation is

- A) doubled.
- B) the same.
- C) decreased by one.
- D) increased by one.
- E) dependent on the temperature.

Answer: B

3) Which of the following gives the balanced equation for this reaction?



- A)  $\text{KPO}_4 + \text{CaNO}_3 + \text{KNO}_3$
- B)  $\text{K}_3\text{PO}_4 + \text{Ca}(\text{NO}_3)_2 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + 3\text{KNO}_3$
- C)  $2\text{K}_3\text{PO}_4 + \text{Ca}(\text{NO}_3)_2 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + 6\text{KNO}_3$
- D)  $2\text{K}_3\text{PO}_4 + 3\text{Ca}(\text{NO}_3)_2 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + 6\text{KNO}_3$
- E)  $\text{K}_3\text{PO}_4 + \text{Ca}(\text{NO}_3)_2 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + \text{KNO}_3$

Answer: D

4) Which of the following correctly gives the best coefficients for the reaction below?



- A) 1, 1, 1, 1
- B) 1, 2, 1, 4
- C) 2, 4, 2, 8
- D) 1, 4, 1, 4
- E) 2, 4, 2, 4

Answer: B

5) What coefficient is placed in front of  $\text{O}_2$  to complete the balancing of the following equation?



- A) 1
- B) 3
- C) 5
- D) 7
- E) 9

Answer: D

6) What is the coefficient of hydrogen,  $\text{H}_2$ , when the following equation is balanced?



- A) 1
- B) 2
- C) 3
- D) 4
- E) 5

Answer: C

7) In this reaction, what is the coefficient for calcium oxide?  $\text{CaO}(\text{s}) + \text{CO}_2(\text{g}) \rightarrow$

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