

POGIL: Periodic Table Trends

(Adapted from Rush Henrietta CSD)

Why:

The Periodic Table is one of the greatest inventions in the history of man. It allows scientists to predict physical and chemical properties of the elements. Dimitri Mendeleev (a Russian scientist) and Robert Mosley (a British chemist) put together this table in the late 1800's based on properties of the elements known at the time. The trends in these properties as you go across periods and down groups is the subject of this Chem POGIL.

Periodic Table of the Elements

Legend:

- Alkali metals
- Alkaline earth metals
- Transition metals
- Nonmetals
- Metals
- Other metals
- Other nonmetals
- Other metalloids

Success Criteria:

- Understand the meaning of *atomic radius, reactivity, electronegativity, ionization energy*
- Recognize trends in atomic radius, reactivity, ionization energy and electronegativity as you go across periods and down groups.

Resources:

Reference Table S, Periodic Table, Unit 3 (Chapter 6) Notes, & Chapter 6 in textbook

Prerequisites:

Answer the following questions using the resources above.

- What happens to the number of valence electrons as you go down a group on the periodic table?
- What happens to the number of valence electrons as you go from left to right across a period?
- What happens to the number of energy levels as you go down a group on the periodic table?
- What happens to the number of energy levels as you go from left to right across a period?
- Draw a Lewis structure for a Cl atom.
- Draw a Bohr diagram for the Cl atom. (Draw above!)
- In this lab we will be talking about atomic radius.
 - Look up the atomic radius of Cl on Table S: _____
 - Why is it possible to show the atomic radius using a Bohr diagram but not by using a Lewis structure?
 - On your Bohr diagram draw an arrow to represent the radius of the Cl atom.
- Define *periodic*:
- Define *trend*:

Periodic Table Pogil Answers

Barbara Millis



Periodic Table Pogil Answers:

Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes

Cooperative Learning in Higher Education Barbara Millis, 2023-07-03 Research has identified cooperative learning as one of the ten High Impact Practices that improve student learning If you ve been interested in cooperative learning but wondered how it would work in your discipline this book provides the necessary theory and a wide range of concrete examples Experienced users of cooperative learning demonstrate how they use it in settings as varied as a developmental mathematics course at a community college and graduate courses in history and the sciences and how it works in small and large classes as well as in hybrid and online environments The authors describe the application of cooperative learning in biology economics educational psychology financial accounting general chemistry and literature at remedial introductory and graduate levels The chapters showcase cooperative learning in action at the same time introducing the reader to major principles such as individual accountability positive interdependence heterogeneous teams group processing and social or leadership skills The authors build upon and cross reference each others chapters describing particular methods and activities in detail They explain how and why they may differ about specific practices while exemplifying reflective approaches to teaching that never fail to address important assessment issues

Atoms, Molecules & Elements: Patterns In the Periodic Table Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice Patterns In the Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

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Atoms, Molecules & Elements Gr. 5-8 George Graybill, 2007-09-01 Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource makes the periodic table easier to understand Begin by answering what are atoms See how the atomic model is made up of electrons protons and neutrons Find out what a molecule is and how they differ from elements Then move on to compounds Find the elements that make up different compounds Get comfortable with the periodic table by recognizing each element as part of a group Examine how patterns in the period table dictate how those elements react with others Finally explore the three important kinds of elements metals nonmetals and inert gases Aligned to the Next Generation Science Standards and written to Bloom s Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included

Periodic Table & Periodic Properties M. Satake, 2010 The book is primarily meant for undergraduate students of chemistry General reader who is interested in chemistry of elements and their behaviour will find it equally interesting and easy to understand

The Periodic Table Desmond Grosvenor Cooper, 1968

The Periodic Table DK, 2025-09-04 This is the periodic table as you ve never seen it before By presenting the elements in the order in which they were discovered The Periodic Table also tells the story of chemistry Now iconic the table was invented in the 1800s long before anyone knew what was inside an atom As more elements were found the table got bigger Some were discovered at risk to life and limb for example the hunt for the explosive element fluorine injured or even killed several scientists who came to be known as the fluorine martyrs Even now the table is still growing as scientists push the boundaries and add to the 118 elements discovered so far Some of the newest elements have been observed only as they briefly flicker into existence in the extreme conditions inside particle colliders and accelerators Gorgeous photographs show even the most familiar elements in a new light Bold clear infographics and artworks reveal aspects of their atomic structure that explain why they behave the way they do For example why is gold so malleable and diamond so incredibly hard How can bismuth levitate and why does cobalt make paint such a vivid blue If you want to unlock the secrets of matter itself open The Periodic Table

The HISTORY of the ATOM the PERIODIC TABLE and RADIOACTIVITY Alex Kimmons, 2021-05-15 THE HISTORY OF THE ATOM THE PERIODIC TABLE and RADIOACTIVITY monochrome Chemistry is an interesting and fundamental branch of science because it gives us the chance to explain the secrets of nature What is water What do we use in our cars as fuel What is aspirin What are perfumes made of These kinds of questions and their answers are all part of the world of chemistry There

is no industry that does not depend upon chemical substances the petroleum pharmaceuticals garment aircraft steel and electronics industries for example as well as agriculture all utilize the science of chemistry This book helps everyone to understand nature However one does not need to be a chemist or scientist to understand the simplicity within the complexity around us The aim was to write a modern up to date book where students and teachers can get concise information about the structure of substances Sometimes reactions are given in the detailed form but in general excessive detail has been omitted The book is designed to introduce fundamental knowledge in three areas the history of the atom the periodic table and radioactivity We will study the historical development of atomic structure theories the tendencies of elements in periods and groups and the types of emissions and uses of radioactivity Elements and the Periodic Table, Grades 5 - 12 Theodore S. Abbigy, 2013-01-02 Aligned to Common Core State Standards Elements and the Periodic Table present the basics of the Periodic Table in an easy to understand easy to master way It contains fun activities transparency masters quizzes tests rubrics grading sheets and more From basic elements to table organization Elements and the Periodic Table is the essential handbook for middle school science Mastering the Periodic Table Linda Trombley, Faye Williams, 2000 Whether students are studying chemistry biology or other sciences the periodic table is a vitally important tool These 50 word games puzzles and other creative activities unlock the nature of the various elements while explicating periodicity atomic structure element groups and more Complete teacher support includes background information answer keys and materials lists **Atoms, Molecules & Elements: What Are Compounds? Gr. 5-8** George Graybill, 2015-10-01 This is the chapter slice What Are Compounds from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives The Periodic Table Book DK, 2017-03-30 This eye popping encyclopedia takes you on a tour of all the world s elements From argon to zinc each and every one of the 118 chemical elements are explored in dazzling detail With the periodic table celebrating its 150th anniversary in 2019 you ll be in your element as you discover the incredible variety of building blocks that make up our Universe and learn the remarkable ways we now use them More than 1 000 photographs showcase the natural forms of each element and the range of everyday and unusual objects where they can be seen This helps children understand exactly where the different elements have found their place in the world The true science behind

the elements is explained in properties atomic structure and table position This essential book turns the tables on traditional reference and presents the periodic table as never before to appeal to school children today Included is a giant glossy pull out poster perfect to aid classroom discussions or to dazzle on bedroom walls **Elements and the Periodic Table, Grades 5 - 8** Abbigy, 2013-01-02 Aligned to Common Core State Standards Elements and the Periodic Table present the basics of the Periodic Table in an easy to understand easy to master way It contains fun activities transparency masters quizzes tests rubrics grading sheets and more From basic elements to table organization Elements and the Periodic Table is the essential handbook for middle school science **Explanatory Key to the Periodic Chart of the Atoms** Henry D. Hubbard, 1926 **Periodic Table** DK, 2020-01-02 A brand new science ebook for kids joins this hugely successful mini encyclopedia series including lot of information Discover all the major elements of the periodic table arranged in chapters according to their group including alkaline earth metals lanthanides and noble gases Every important element from hydrogen via carbon oxygen and gold to oganesson that s element number 118 is presented here Each element is featured in its own catalogue style entry with a photo of the element in its raw state or in use a caption explaining important information ranging from where it s found to its main uses and a fact list presenting the key data including the date of discovery and all atomic information as found on the periodic table Plus there are stunning full page photos showing elements as you have never seen them before perhaps inside machinery that you can t normally delve into or in stunning macro photography that reveals microscopic details invisible to the human eye The style of the Pocket Eyewitness series is perfect for all children from reluctant readers who can easily digest the key points through to budding Marie Curies and Louis Pasteurs who want to know more about the most essential particles on the planet Chemical Elements Pocket Guide Coventry House Publishing, 2020-02-08 The Chemical Elements Pocket Guide serves as a portable reference for quick study and efficient review of the 118 elements on the periodic table This on the go resource details the physical and atomic properties of each element as well as their history and characteristics in bullet point format The book s small trim size 4 25 x 6 8 inches is intended to fit inside a lab coat pocket and the bound design means you no longer need to carry loose bulky flashcards that can be misplaced or destroyed Includes the updated names nihonium moscovium tennessine and oganesson for elements 113 115 117 and 118 respectively Information provided includes Atomic number Atomic symbol Element category Standard state Atomic mass Electron configuration Oxidation states Electronegativity Atomic radius Ionization energy Electron affinity Melting point Boiling point Density Year discovered Discovered by Appearance Natural occurrence Interesting fact

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