

Atomic Structure & Periodicity

1. What did JJ Thomson discover?
a) Proton b) Neutron c) Quark d) Electron
2. Millikan discovered the mass and charge of which particle?
a) Proton b) Neutron c) Electron d) Alpha particle
3. Which of the following types of radiation has the longest wavelength?
a) gamma rays b) visible c) ultraviolet d) radar
4. Which of the following has the longest wavelength?
a) blue light b) yellow light c) red light d) green light
5. Which of the following types of radiation has the highest energy?
a) gamma rays b) visible c) ultraviolet d) radar
6. Which of the following is NOT a proper unit of frequency?
a) Hz b) s^{-1} c) $m \cdot s^{-1}$ d) $1/s$
7. What is the relationship between the energy of a photon of light and its frequency?
a) $E = v$ b) $E = h/v$ c) $E = hv$ d) $E = 1/hv$
8. What is the de Broglie wavelength of an electron moving at 80% of the speed of light?
a) $3.03 \times 10^{-12} \text{ m}$ b) $2.42 \times 10^{-12} \text{ m}$ c) $3.30 \times 10^{-11} \text{ m}$ d) $1.59 \times 10^{-25} \text{ m}$
9. Which quantum number determines the subshell occupied by the electron?
a) n b) l c) m_l d) m_s
10. Which of the following electronic transitions in a hydrogen atom would have the longest wavelength?
a) $n = 4$ to $n = 1$ b) $n = 4$ to $n = 2$ c) $n = 2$ to $n = 1$ d) $n = 4$ to $n = 3$

Periodicity Worksheet Answer Guide

**MrExcel's Holy Macro! Books, Liam
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