

- * 3. A Which of the following best represents the fraction of a radioactive sample that remains after four half-lives have occurred?

(a) $\left(\frac{1}{2}\right)^4$

(c) $\left(\frac{1}{4}\right)$

(b) $\left(\frac{1}{2}\right) \times 4$

(d) $\left(\frac{1}{2}\right)^2 \times 4$

4. Match the nuclear symbol on the right to the name of the corresponding particle on the left.

C beta particle

(a) ${}^1_1\text{P}$

A proton

(b) ${}^4_2\text{He}$

D positron

(c) ${}^0_{-1}\beta$

B alpha particle

(d) ${}^4_2\alpha$

5. Label each of the following statements as True or False. Consider the U-238 decay series in Figure 2.7 of the text. For the series of decays from U-234 to Po-218, each nuclide

F a. shares the same atomic number

F b. differs in mass number from others by multiples of 4

F c. has a unique atomic number

F d. differs in atomic number from others by multiples of 4

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Sabine Zange



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